



O efeito da metila na Química Medicinal

DQ - Universidade de Aveiro – 12 de setembro de 2012

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Professor Titular



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Laboratório de Avaliação e Síntese de Substâncias Bioativas

www.farmacia.ufrj.br/lassbio

Instituto Nacional de Ciência e Tecnologia de Fármacos e Medicamentos
(INCT-INOVAR)



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Universidade Federal do Rio de Janeiro



LASSBIO

Laboratório de Avaliação e Síntese de Substâncias Bioativas

Cidade Universitária, ilha do Fundão,
Rio de Janeiro, BR

Laboratório de Avaliação e Síntese de Substâncias Bioativas

Química Medicinal



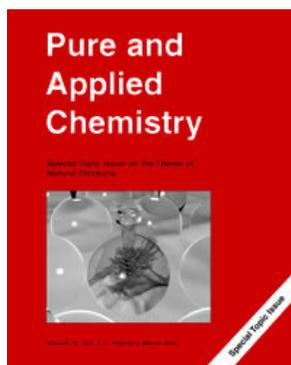
Criado em 19/04/1994



IUPAC - Subcommittee Medicinal Chemistry & Drug Development

Química Medicinal é a *disciplina* que

estuda aspectos **relacionados** à descoberta ou invenção de **fármacos**, seus aspectos moleculares envolvidos no mecanismo de ação e aqueles que governam a **absorção**, **distribuição**, **metabolismo**, **eliminação** e **toxicidade** (ADMET), incluindo a compreensão da relação entre a estrutura química e a atividade terapêutica (REA = *SAR*).

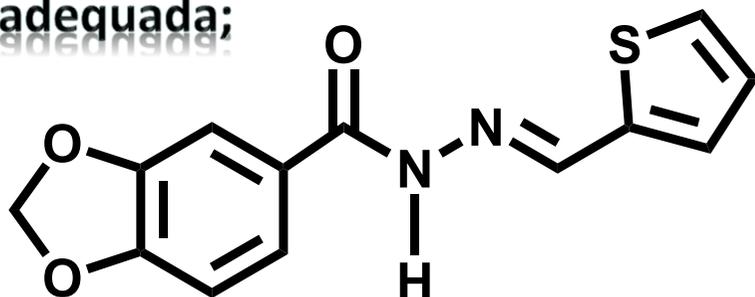


IUPAC

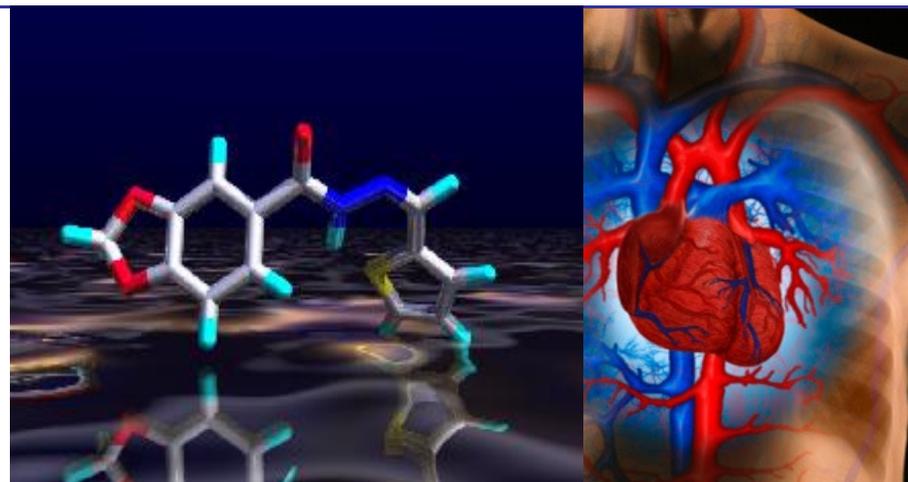
Pure & Appl. Chem., Vol. 70, No. 5, pp. 1129–1143, 1998.
Printed in Great Britain.
© 1998 IUPAC



Derivado estruturalmente simples;
sinteticamente acessível em ótimos
rendimentos e em escala 20 M, a partir
de matéria-prima natural sustentável &
acessível; com estabilidade química e
metabólica; com biodisponibilidade
adequada;



LASSBio-294



Protótipo de fármaco cardioativo,
com potentes propriedades
inotrópicas, vasodilatadoras
& neuroprotetoras;

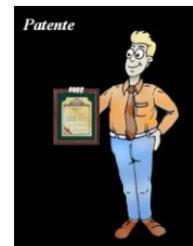
Ativo *p.o.* por novo mecanismo de ação;

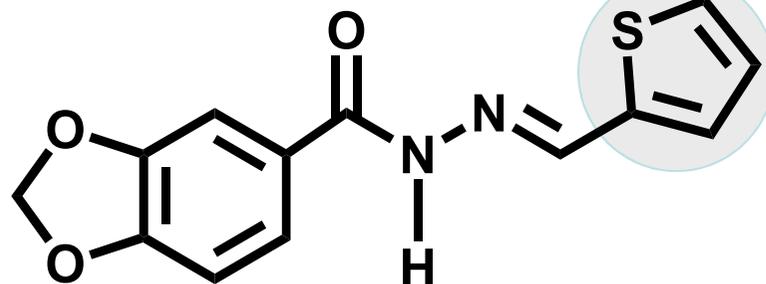
Sem cito-, embrio- & genotoxicidade.

Possíveis indicações terapêuticas

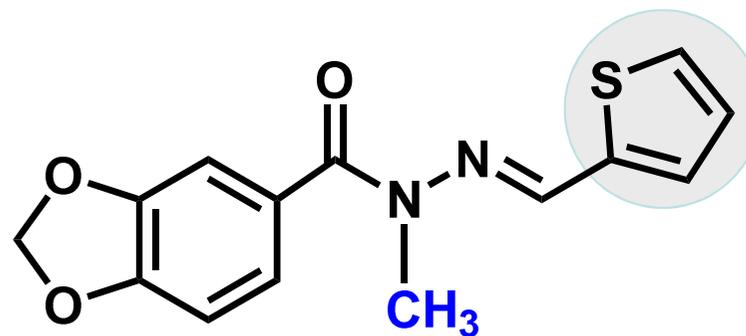
Cardiopatias c/ hipertensão

US Pat. 7,091,238 (15 de agosto de 2006)

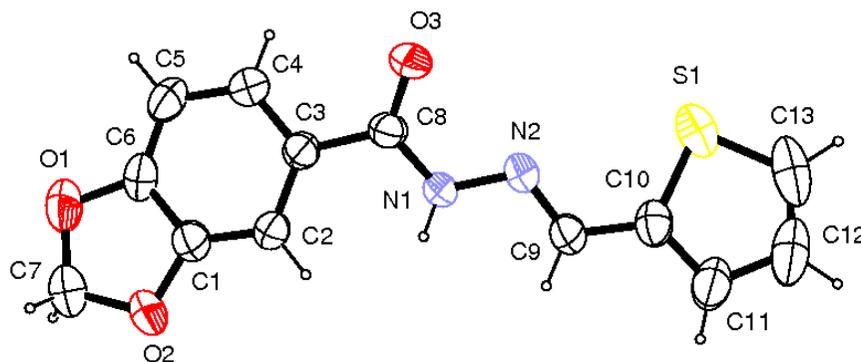




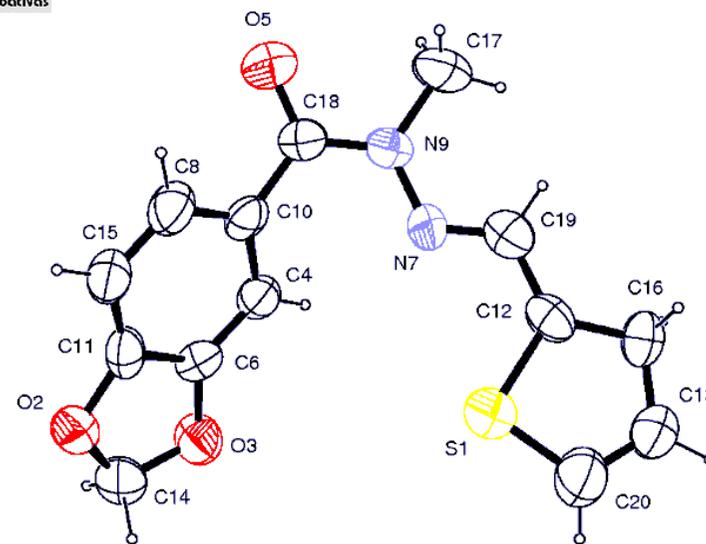
LASSBio-294



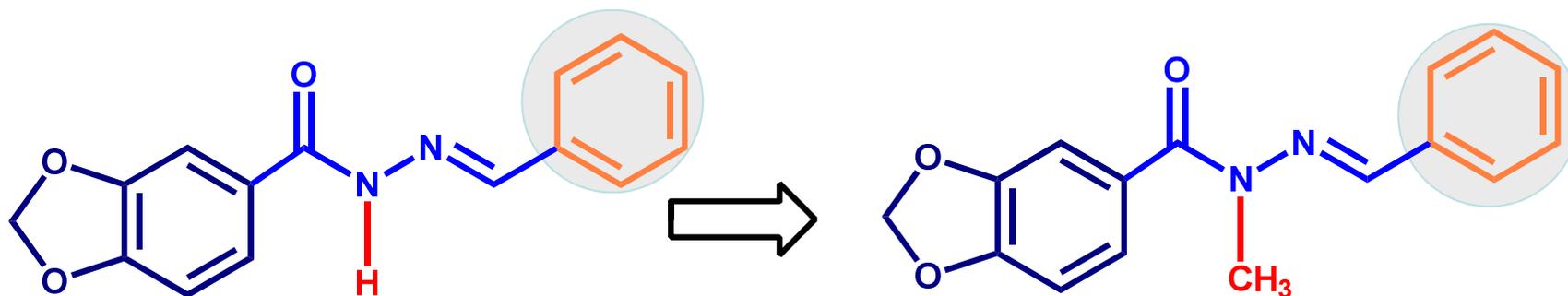
LASSBio-785



Conformação “grampo-de-cabelo”

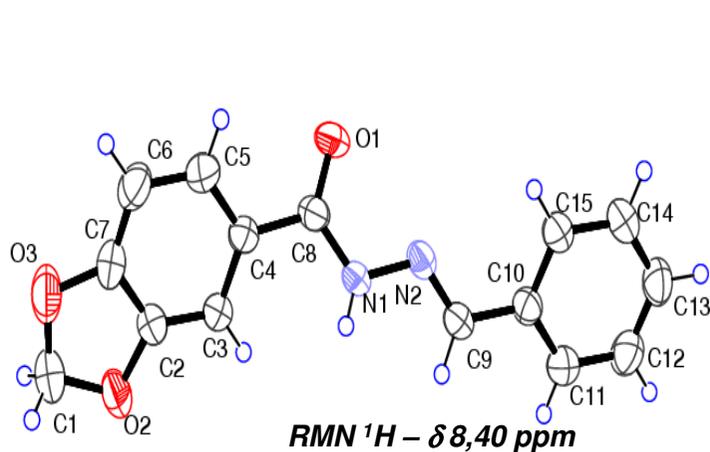


Conformação em “U”

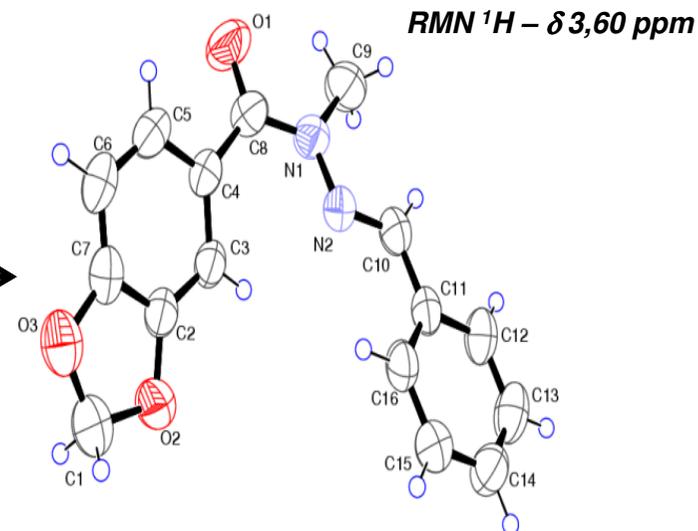


LASSBio-123

LASSBio-1004



LASSBio-123



LASSBio-1004

Conformação “grampo-de-cabelo”

Conformação em “U”



15 Da

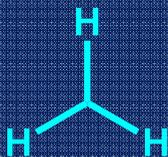
C-H $\mu = 0,4 \text{ D}$

δ^+ / R^+

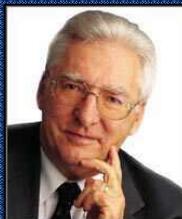
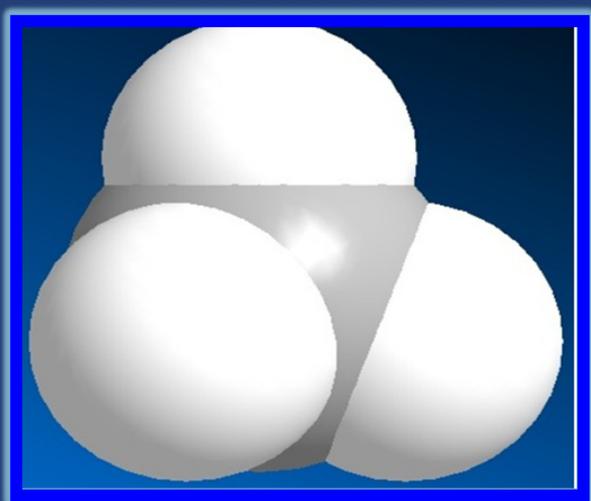
$P = 0,22$

$\sigma_{\text{meta}} = 0,51 / \sigma_{\text{para}} = 0,52$

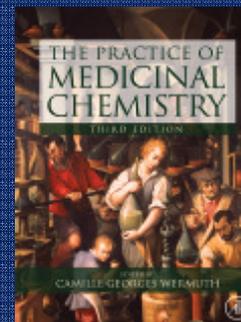
Rekker const = 0,702



“The methyl group, so often considered as chemically inert, is able to alter deeply the pharmacological properties of a molecule.”



Camille G. Wermuth





CHEMICAL REVIEWS

Chem. Rev. 2011, 111, 5215–5246

REVIEW

pubs.acs.org/CR

The Methylation Effect in Medicinal Chemistry

Eliezer J. Barreiro,^{*,†,‡,§} Arthur E. Kümmerle,^{||,†,§} and Carlos A. M. Fraga^{†,‡,§}

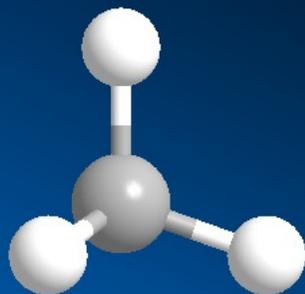
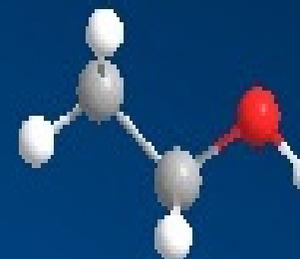
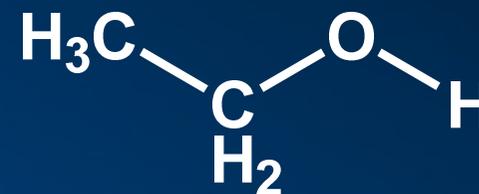
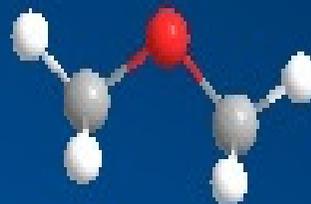
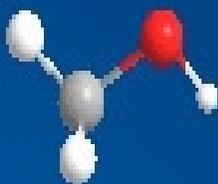
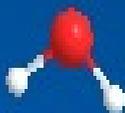
[†]Laboratório de Avaliação e Síntese de Substâncias Bioativas (LASSBio), Faculdade de Farmácia, Universidade Federal do Rio de Janeiro, CCS, Cidade Universitária, CP 68.006, 21941-902 Rio de Janeiro, RJ, Brazil

[‡]Programa de Pós-Graduação em Farmacologia e Química Medicinal, Instituto de Ciências Biomédicas, Universidade Federal do Rio de Janeiro, Cidade Universitária, Ilha do Fundão, Rio de Janeiro, RJ, Brazil

[§]Programa de Pós-Graduação em Química, Instituto de Química, Universidade Federal do Rio de Janeiro, Cidade Universitária, Ilha do Fundão, Rio de Janeiro, RJ, Brazil

dx.doi.org/10.1021/cr200060g

A homologia do carbono...

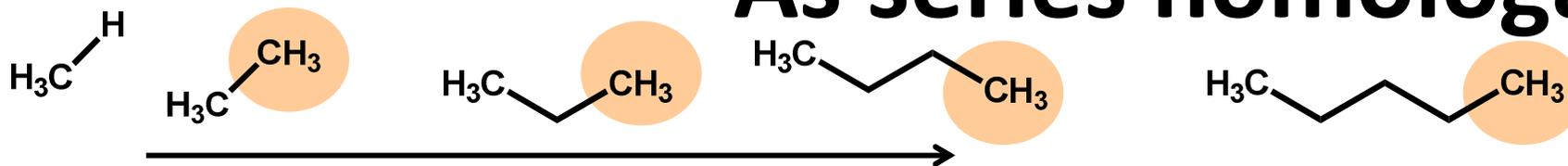


metanol

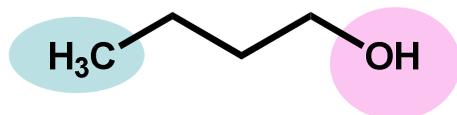
etanol



As séries homólogas

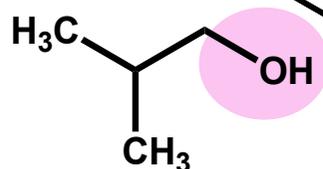


Lipofilicidade



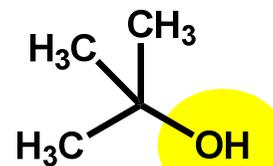
n-butanol

solubility in water
8.2g/100g



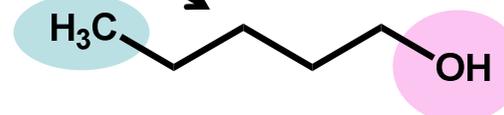
isobutanol

solubility in water
12,5 g/100g



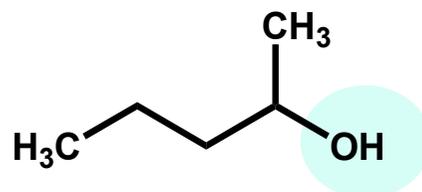
tert-butanol

solubility in water
miscible



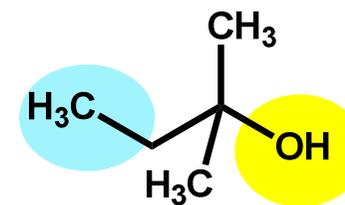
n-pentanol

solubility in water
2.4g/100g



2-pentanol

solubility in water
4.9g/100g



neopentanol

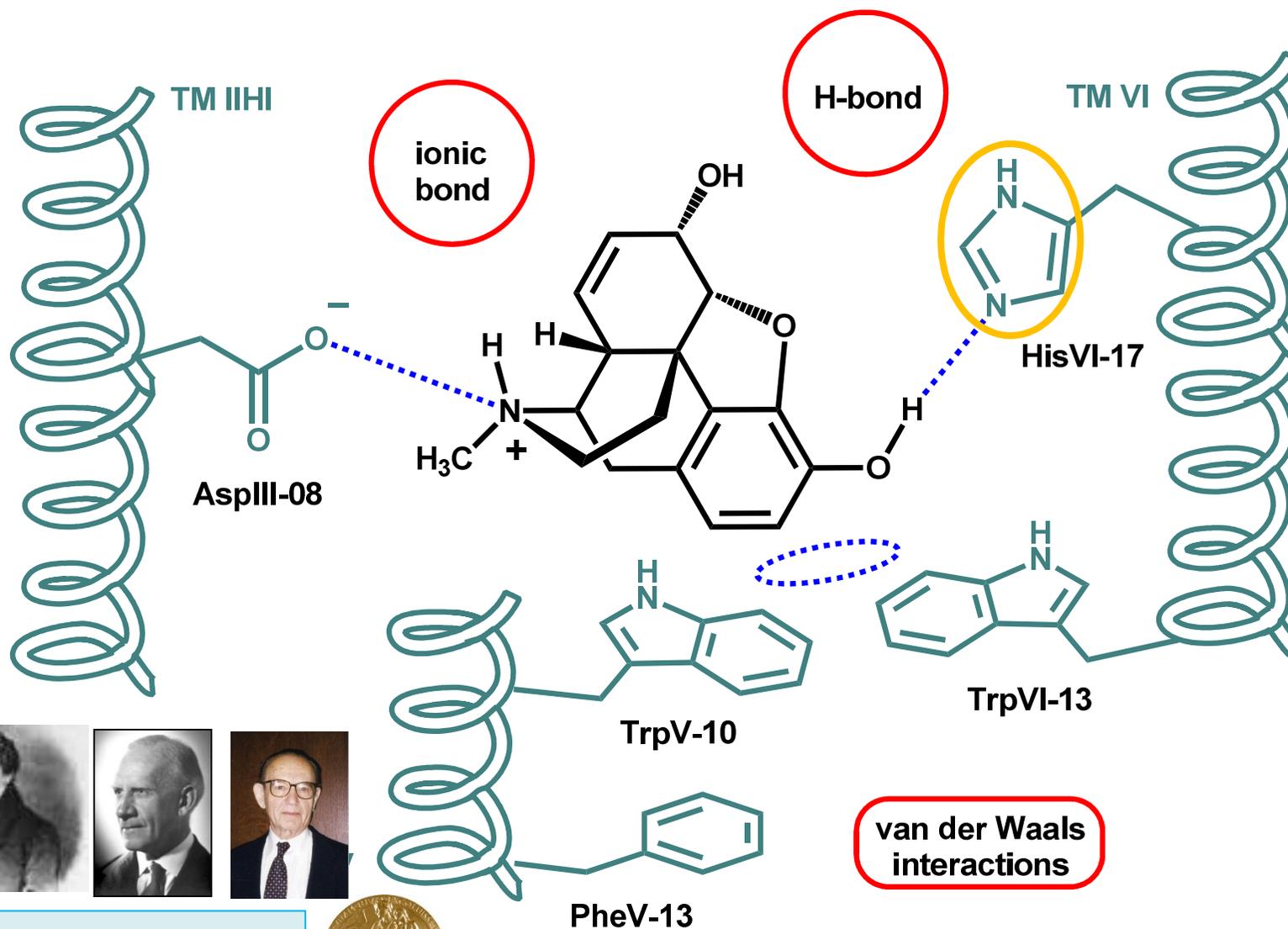
solubility in water
12.2g/100g

G Némethy, *Angew Chem Int Ed Engl* 1967, 6, 195





A metila e a morfina...



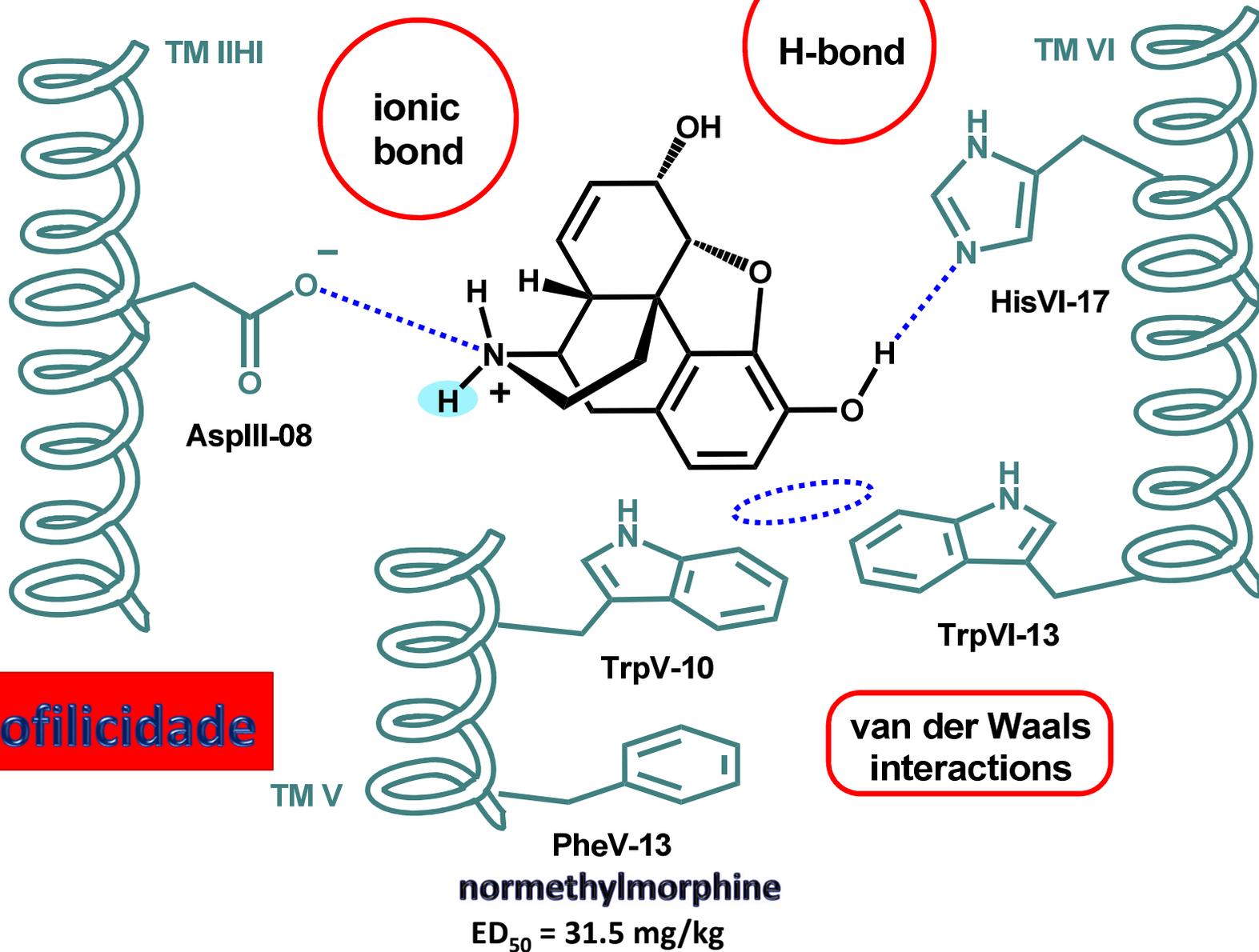
1805 - F. Setürner
1925 - R. Robinson
2001 - G. Stork



1947

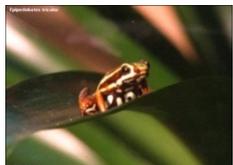
Morphine
ED₅₀ = 4.8 mg/kg

A metila e a N-normetilmorfina...

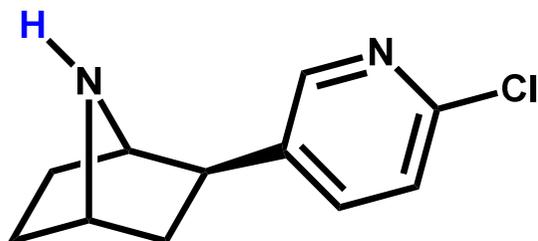




O charme da metila na epibatidina...

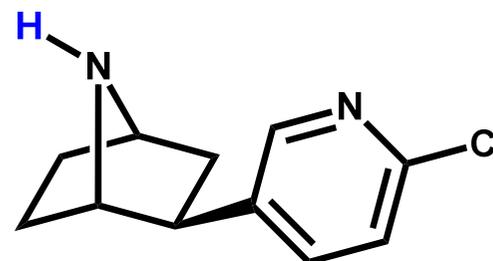


Epipedobates tricolor

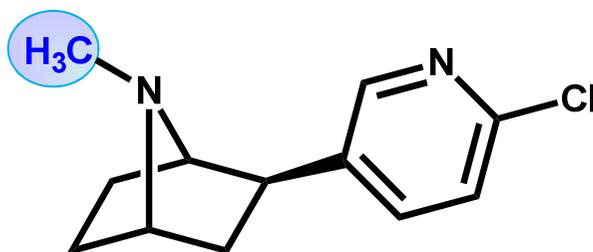


(-)-epibatidina (natural)

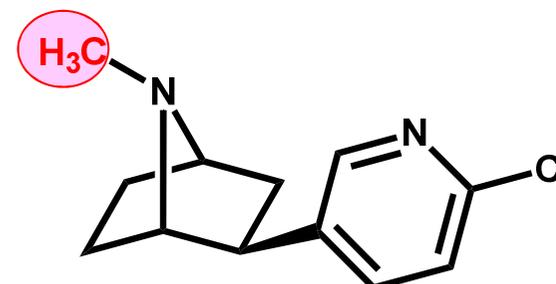
nAChR



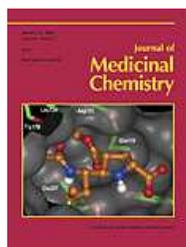
(+)-epibatidina



(-)-*N*-metilepibatidina (natural)



(+)-*N*-metilepibatidina



J W Daly, "Ernest Guenther Award in Chemistry of Natural Products. Amphibian Skin: A Remarkable Source of Biologically Active Arthropod Alkaloids", *J. Med. Chem.* 2003, 46, 445-452



J W Daly, "Thirty Years of Discovering Arthropod Alkaloids in Amphibian Skin", *J. Nat. Prod.* 1998, 61, 162-172



John W. Daly
Un. Maryland, Baltimore, EUA



National Historic Chemical Landmarks

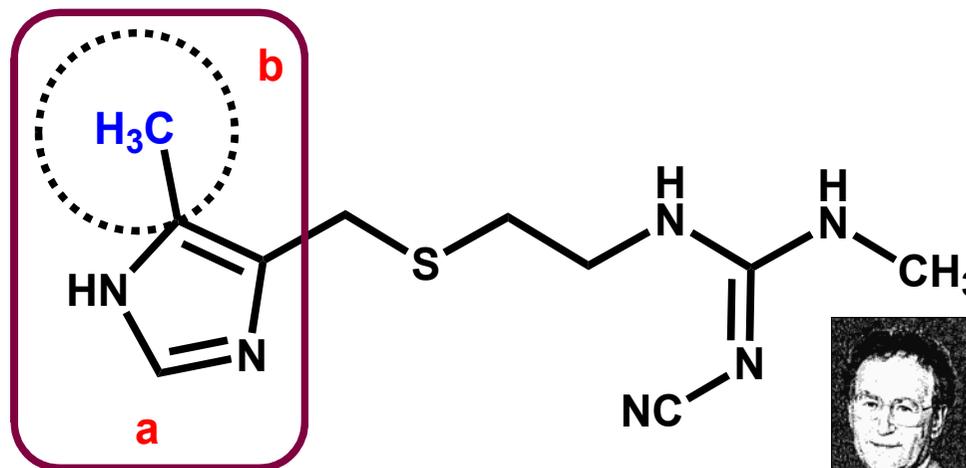
AMERICAN CHEMICAL SOCIETY

A new era of logical drug design

The research program leading to cimetidine also represented a revolution in the way pharmaceuticals are developed. Traditionally, the development of a new drug would often depend on the fortuitous discovery of a plant or microbial extract that showed some of the required biological activity. Using that first extract as a lead, many similar compounds would be made and tested for pharmacological effectiveness. In many cases, the researchers did not know how the drug worked, so finding an optimal compound was difficult.

The development of cimetidine was radically different: it was one of the first drugs to be designed logically from first principles. SK&F's multidisciplinary research team first looked at the physiological cause of acid secretion. They confirmed that a molecule found in the body called histamine triggers the release of acid when it binds to a specific receptor (now called the H₂-receptor) in the stomach lining. Their aim was to find a molecule that successfully competed with histamine in combining with the receptor, but then blocked, rather than stimulated, acid release. Such a molecule was called a histamine H₂-receptor antagonist and represented a new class of drugs.

H₂-receptor antagonist

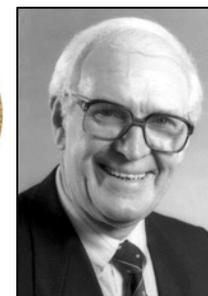


cimetidine

Inovação terapêutica



1988



James W. Black



C Robin Ganellin



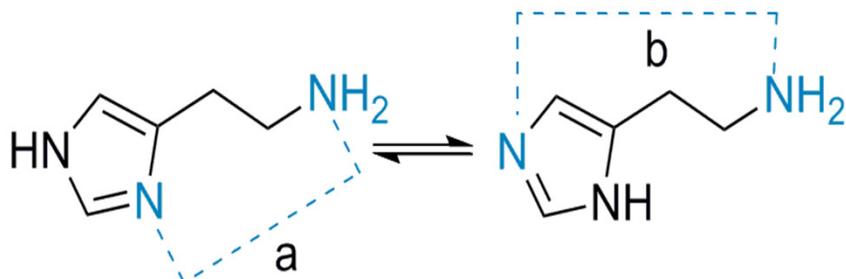
John C Emmett



Graham J Durant

Dois sub-tipos de H_R

Interações fracas

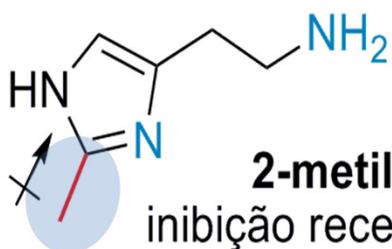


Forma A

$a = 4,83 \text{ \AA}$
 $b = 5,52 \text{ \AA}$

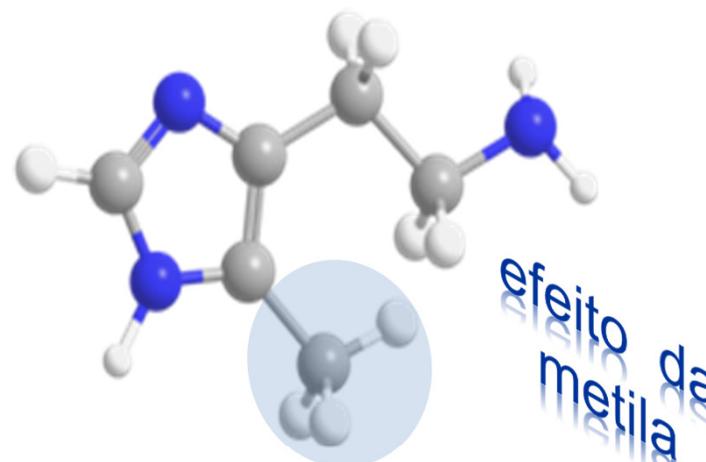
Forma B

Equilíbrio tautomérico

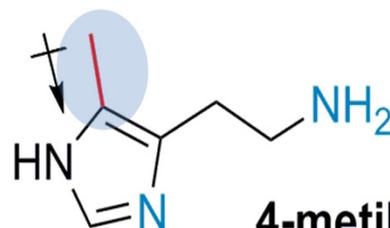


2-metil-histamina

inibição receptores H₁ = 17%
inibição receptores H₂ = < 2%



4-metil-histamina



Análogo ativo

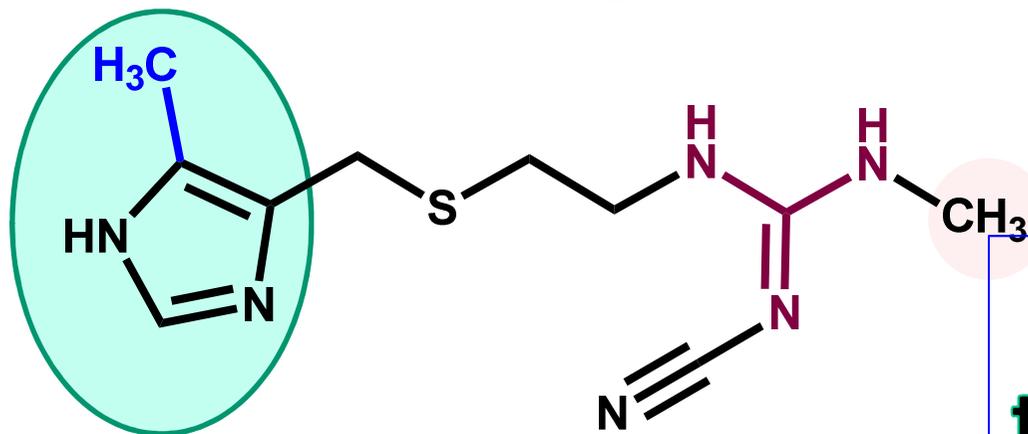
4-metil-histamina

inibição receptores H₁ = 0,2%
inibição receptores H₂ = 50%

CR Ganellin, *Drug Discov Today* 2004, 9, 158

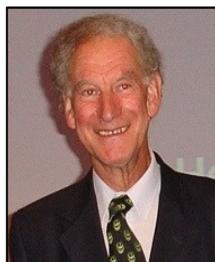


A metila e a invenção da cimetidina



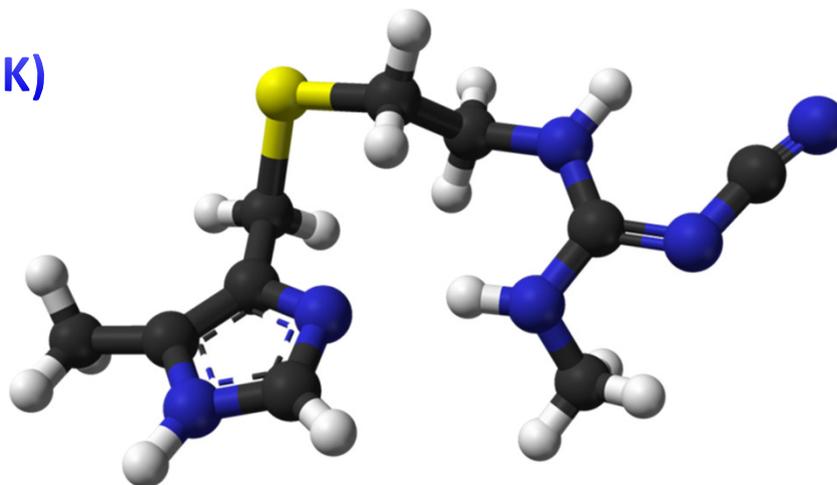
**Inovação
terapêutica**

Smith, Kline and French
(SK&F)
(atual GSK)



C Robin Ganellin

cimetidina
 $C_{10}H_{16}N_6S$



1st blockbuster

1975

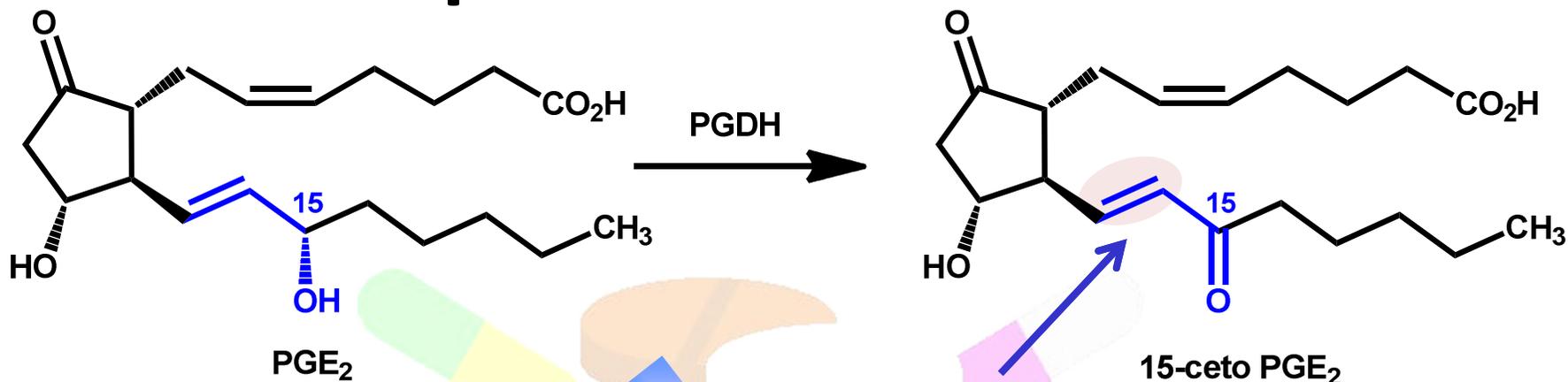
> US\$ 1 bi



Antagonista seletivo de receptor histaminérgico H₂

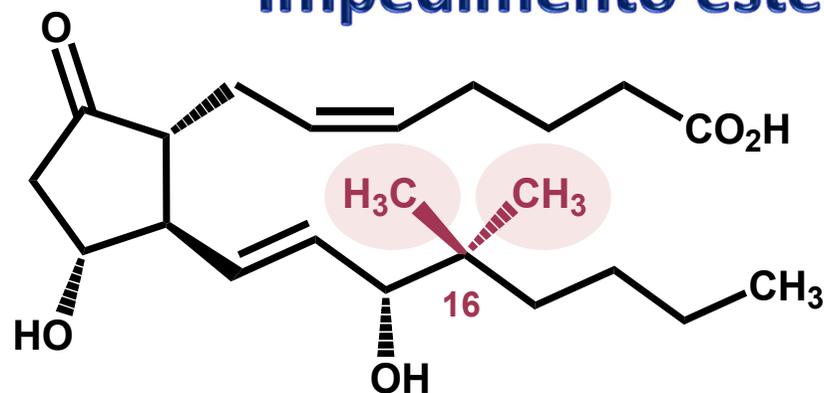


A metila previnando o metabolismo



PG-redutase (PGR)

Impedimento estérico



EP₂ ag (1 nM)



John R. Vane
(1927-2004)



Sune K. Bergström
(1916-2004)

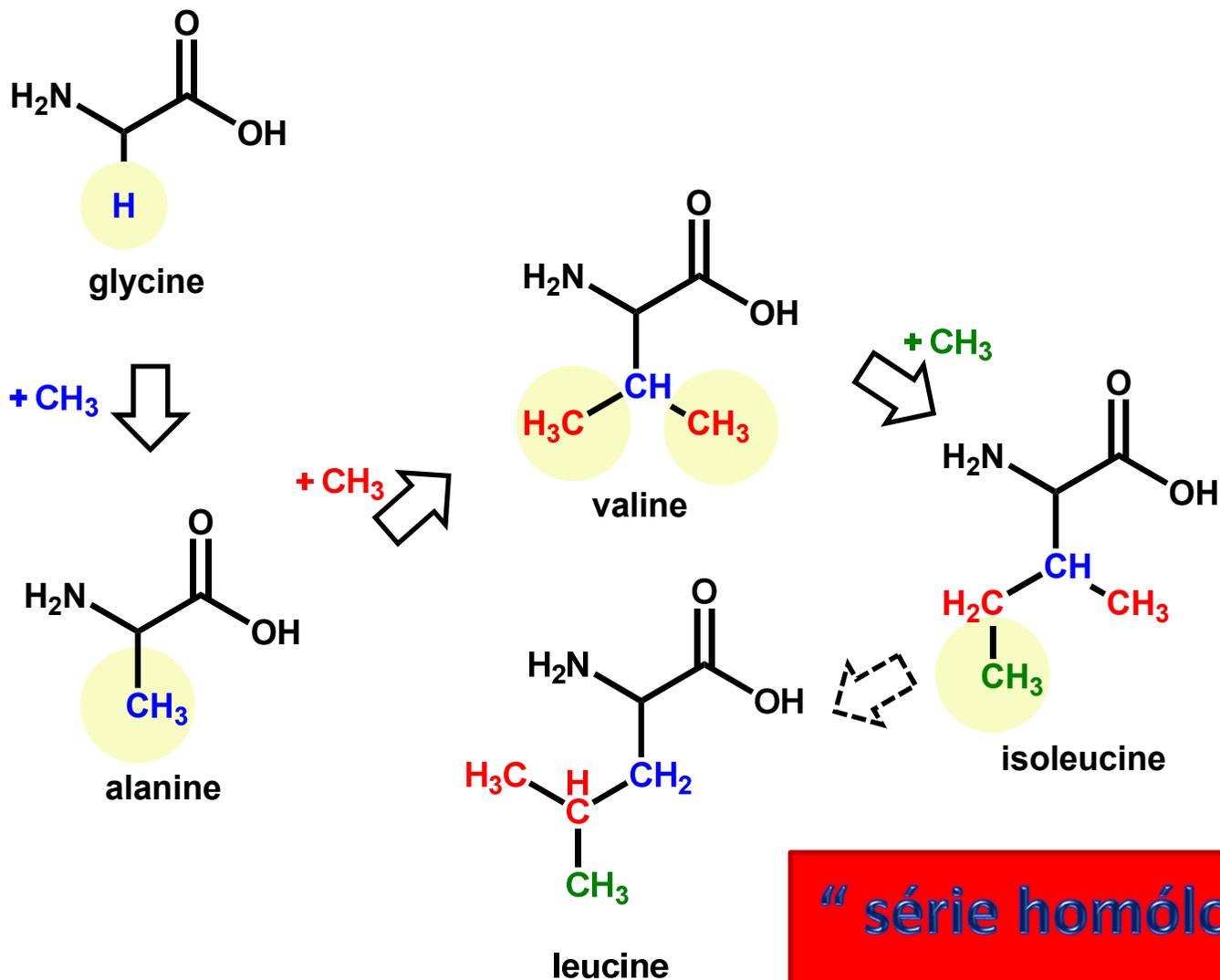


1982



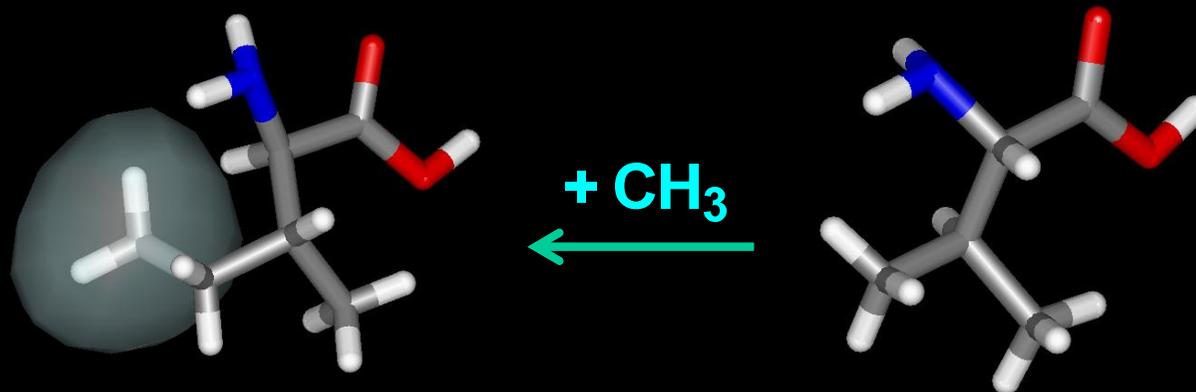
Bengt I. Samuelsson

A homologia e os aminoácidos



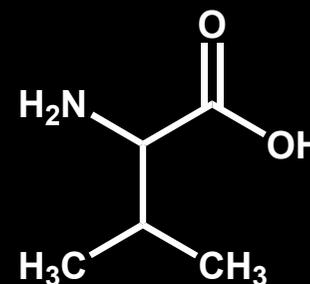
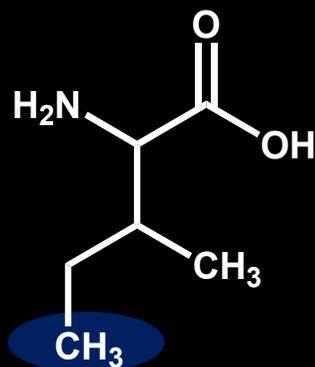
“ série homóloga ”

A homologia e os aminoácidos



isoleucina

valina

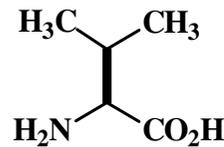


Os amino ácidos homólogos e a COX

COX-2

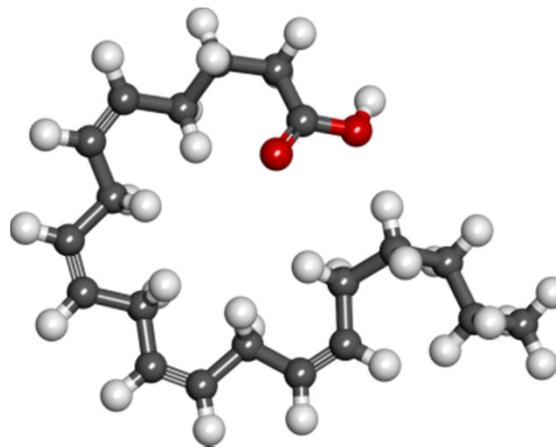
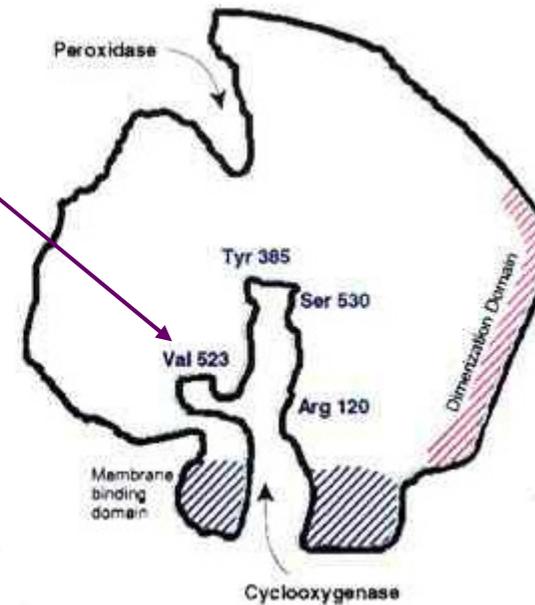
- Inflamação
- Câncer
- Endotélio vascular
- Rins
- Cérebro

Sítio secundário

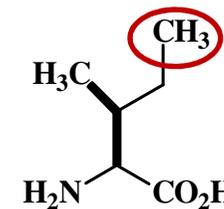
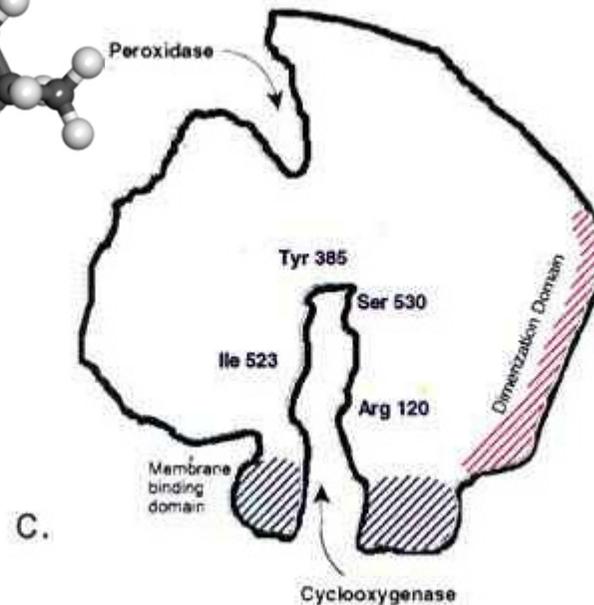


$C_5H_{11}NO_2$
Valina

b.



Ácido araquidônico
 $K_m = 5,6/5,4 \mu M$

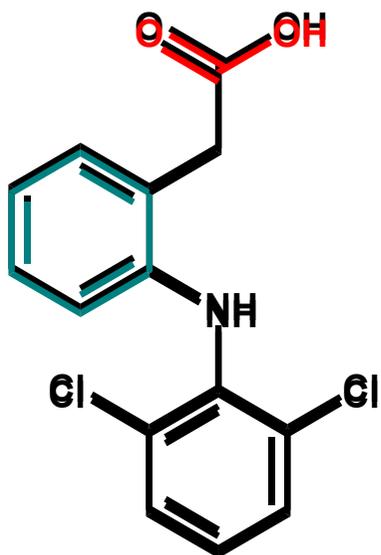


$C_6H_{13}NO_2$
Isoleucina

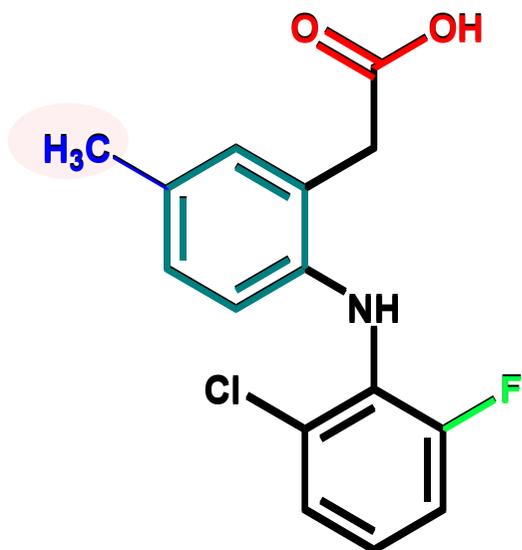
COX-1

- Estômago
- Plaquetas
- Rins

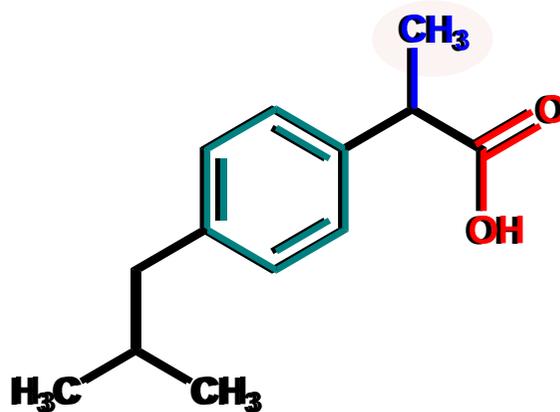
Antiinflamatórios não esteróides



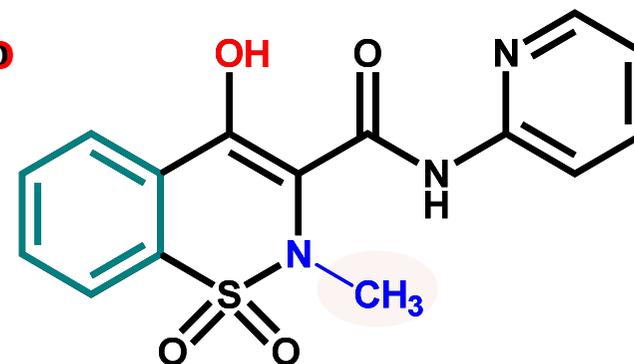
diclofenac



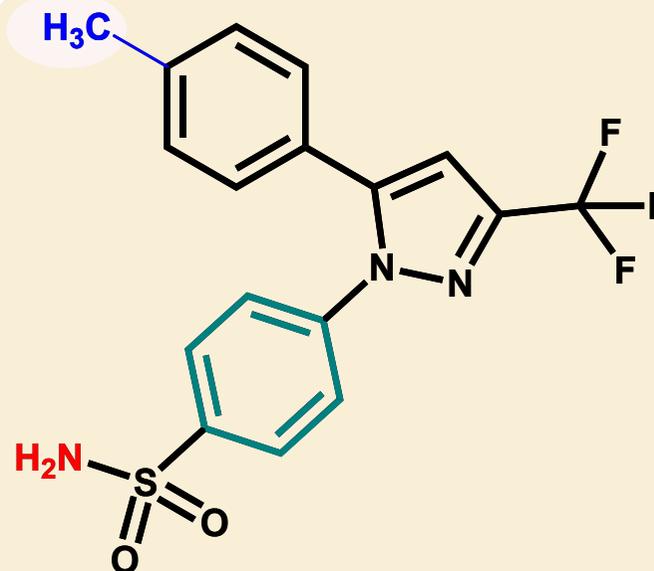
lumiracoxib



ibuprofen

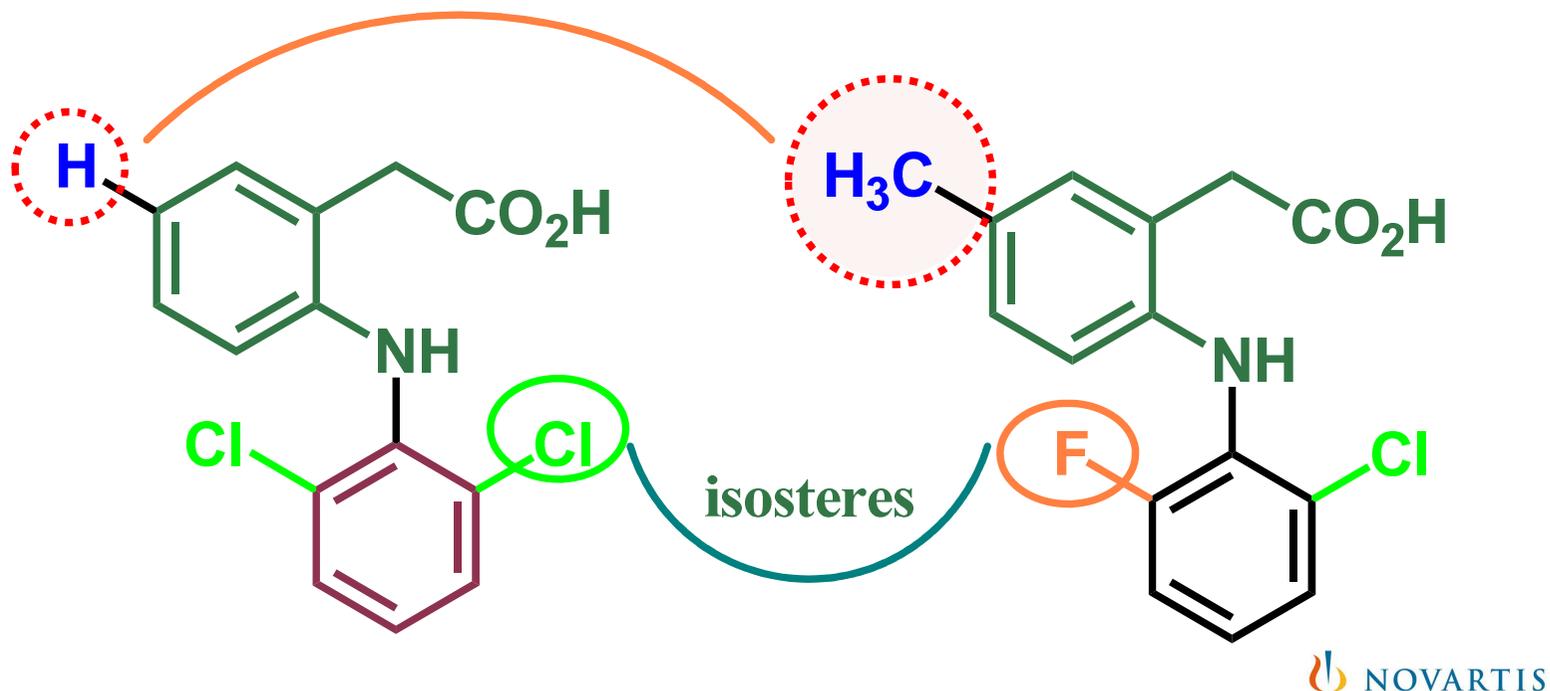


piroxicam



celecoxib

O charme da metila...



DICLOFENAC

COX-1

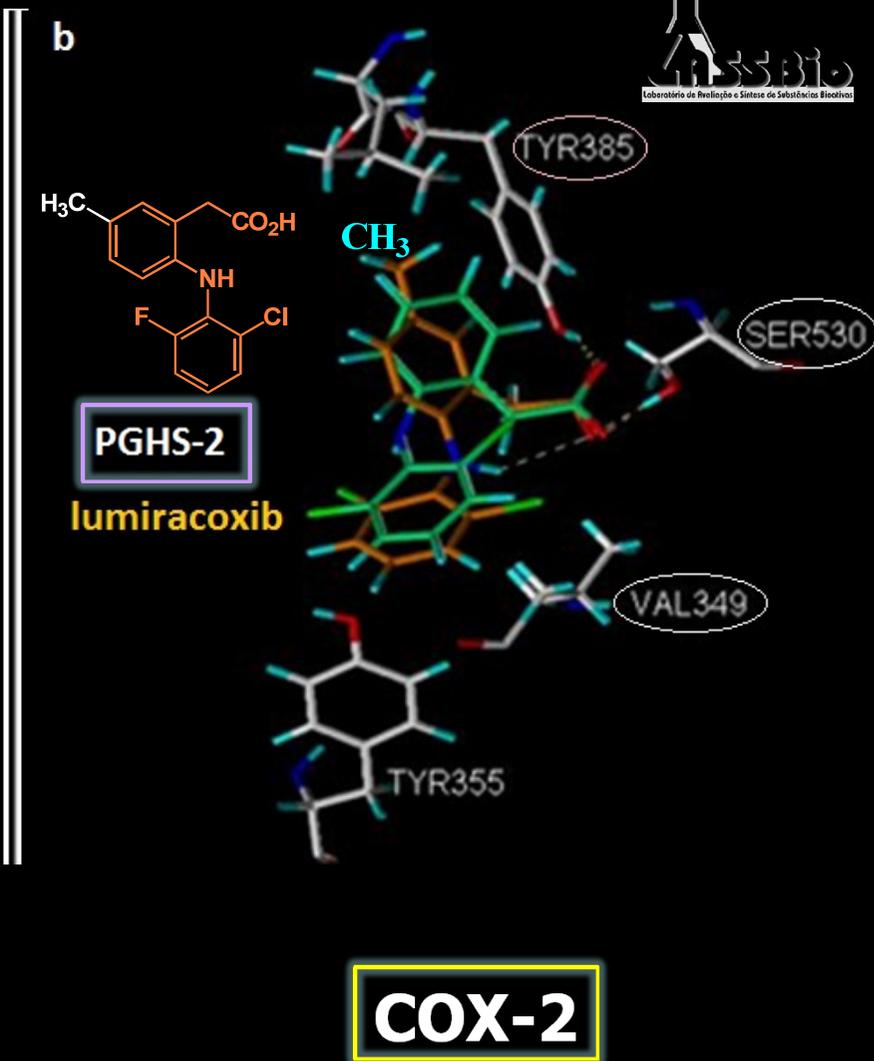
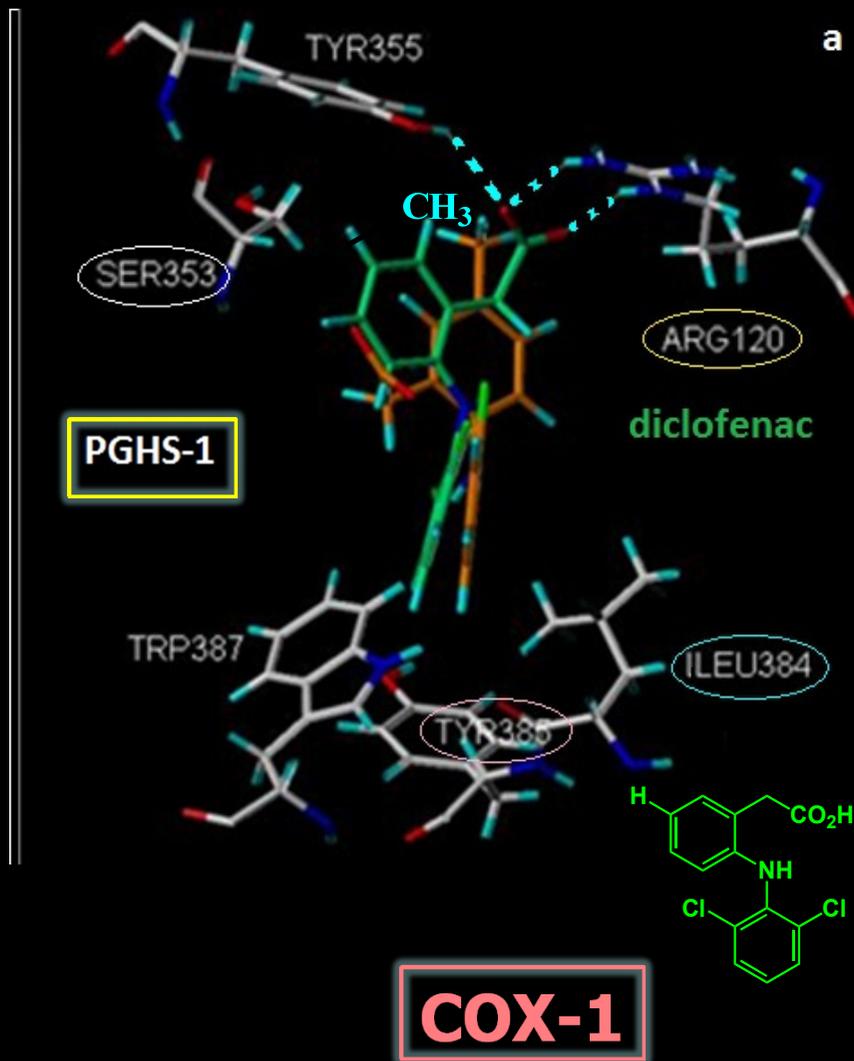
LUMIRACOXIB

COX-2

2006 (2007)

Lumiracoxib have one chlorine substituted by fluorine and the phenylacetic acid moiety has methyl group in *meta* position

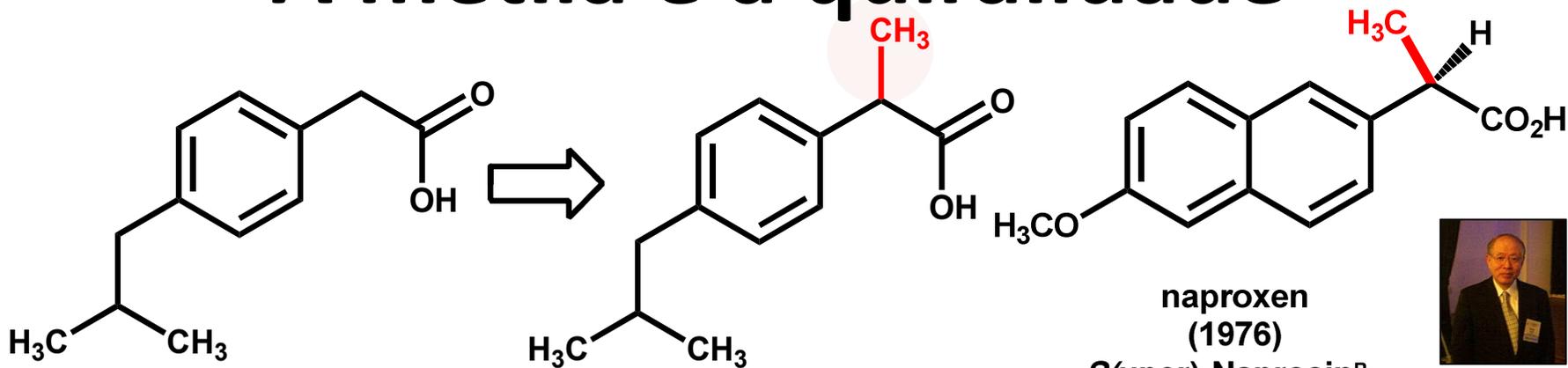
A seletividade entre COX-1 e COX-2



CM Corrêa et al., *Letters in Drug Design & Discovery*, 2007, 4, 422



A metila e a quiralidade



ibufenac

ibuprofen (1961)

naproxen (1976)
S(uper)-Naprosin[®]

IT Harrison, *J Med Chem* 1970, 13, 203



R Noyori

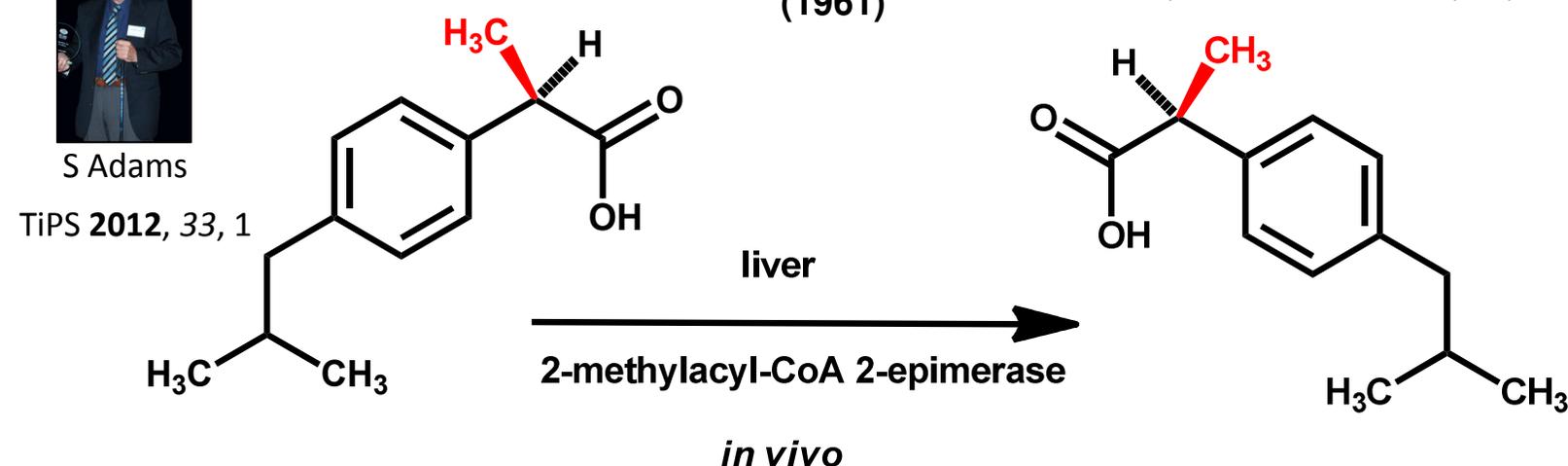


S Adams

TiPS 2012, 33, 1

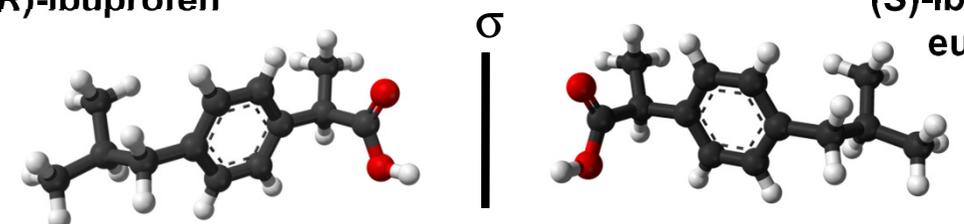


2001



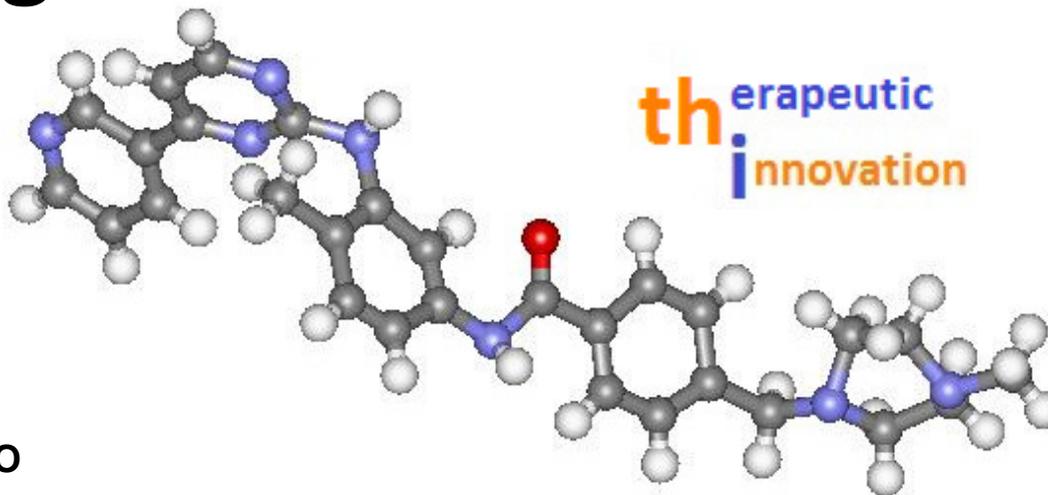
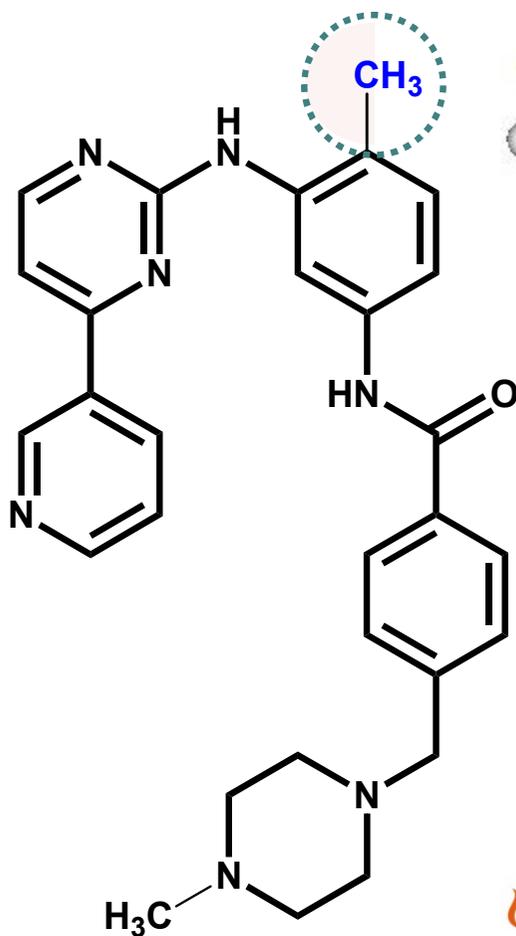
(R)-ibuprofen

(S)-ibuprofen eutômero





A metila e a gênese do imatinibe



therapeutic
innovation



imatinibe
2001



Jürg Zimmermann



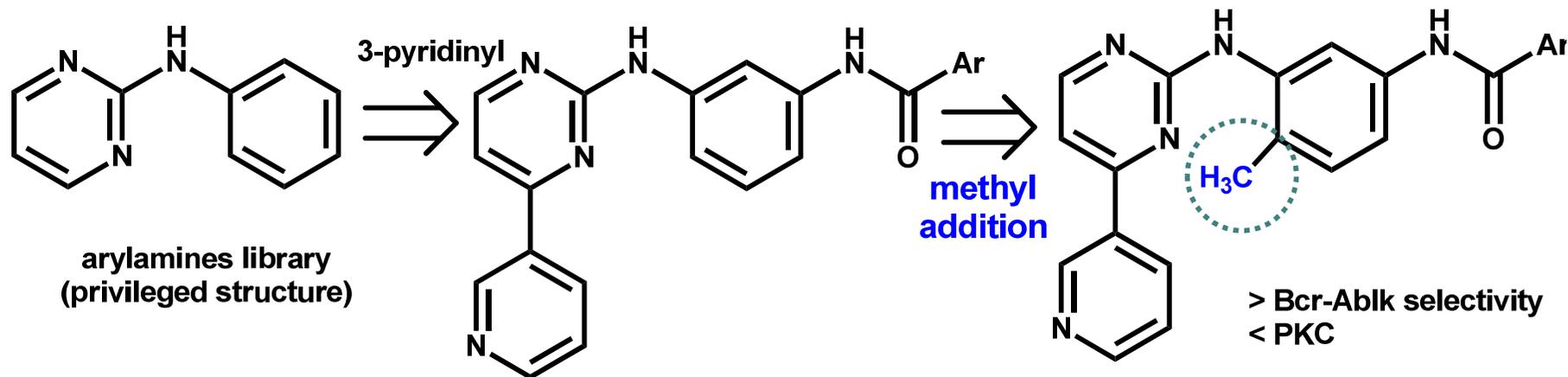
Elisabeth
Buchdunger



Brian J. Druker



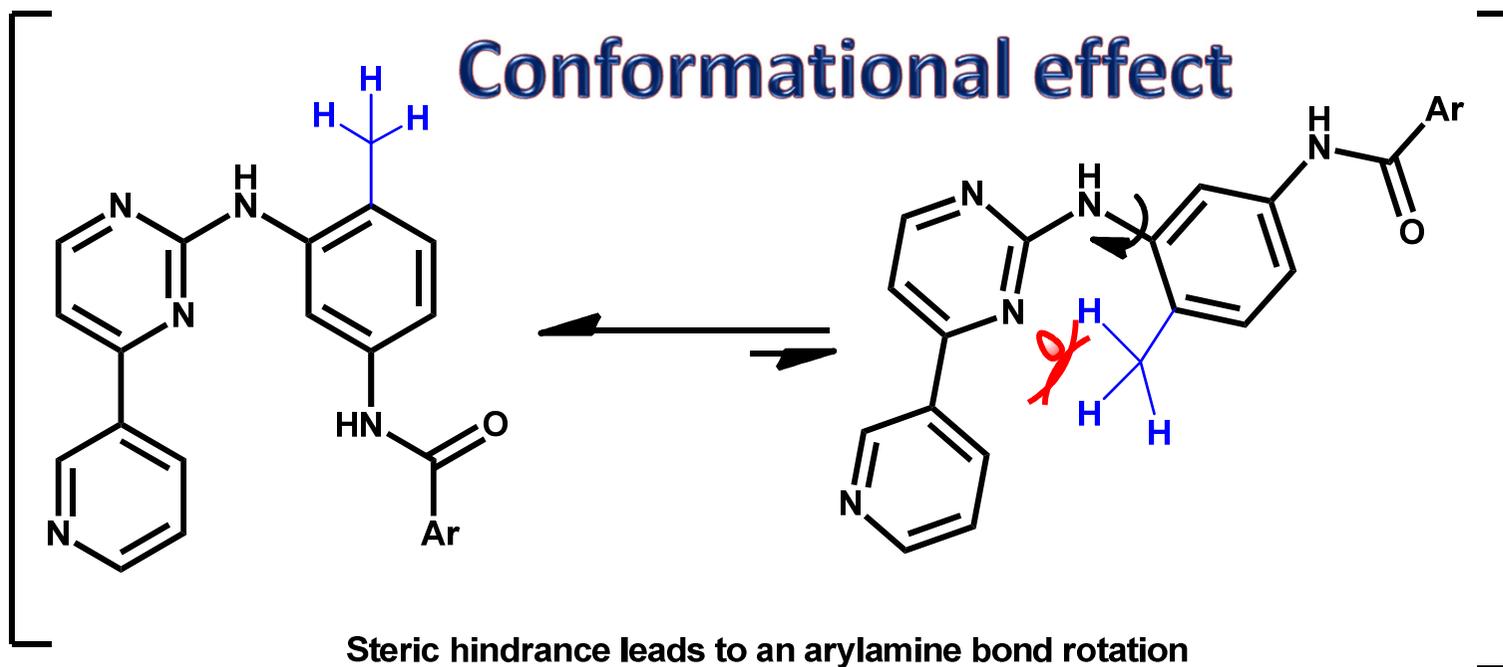
A metila e a gênese do imatinibe

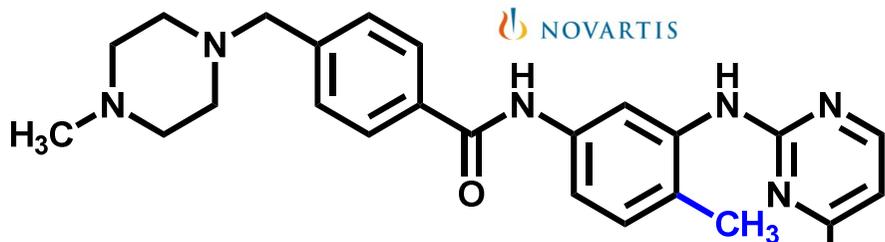


PKC and TK inhibitor
(Bcr-Ablk inhibitor)

> Bcr-Ablk selectivity
< PKC

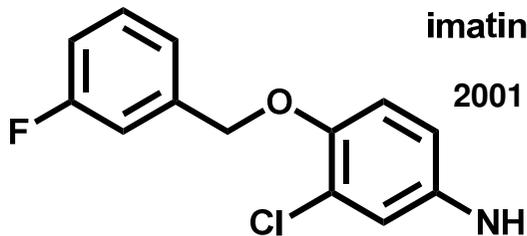
Conformational effect





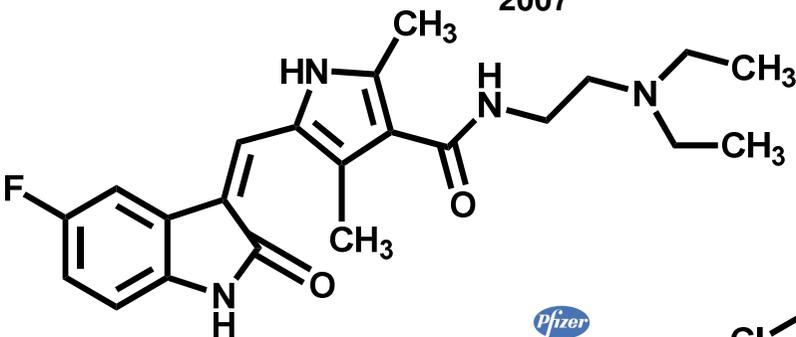
imatinibe

2001



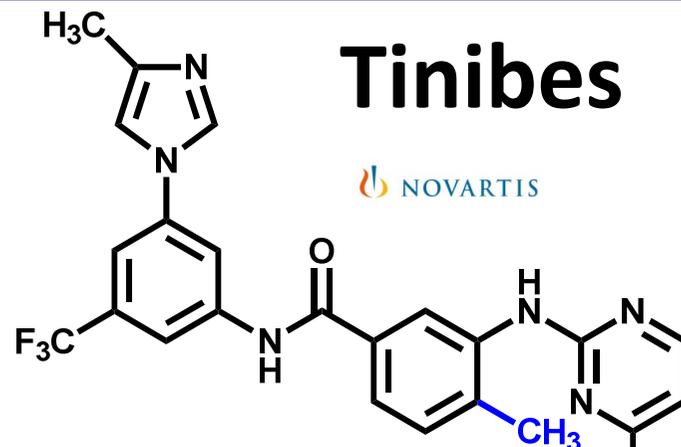
lapatinibe

2007



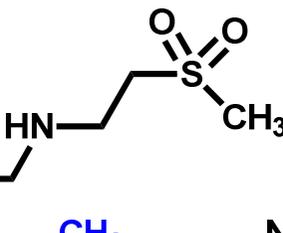
sunitinibe

2006



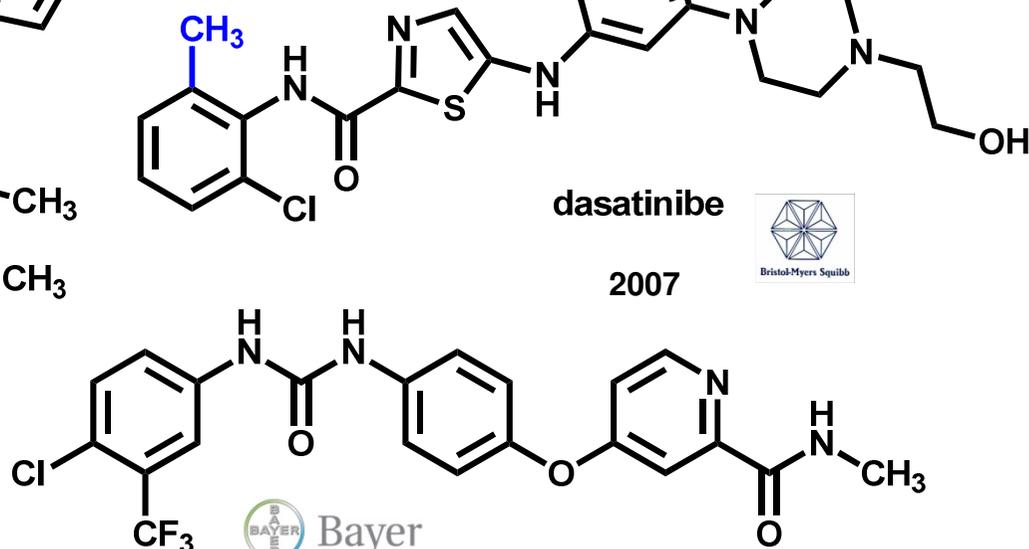
nilotinibe

2006



dasatinibe

2007



sorafenibe

2007



• US market in 2009: US\$ 18,5 bi *



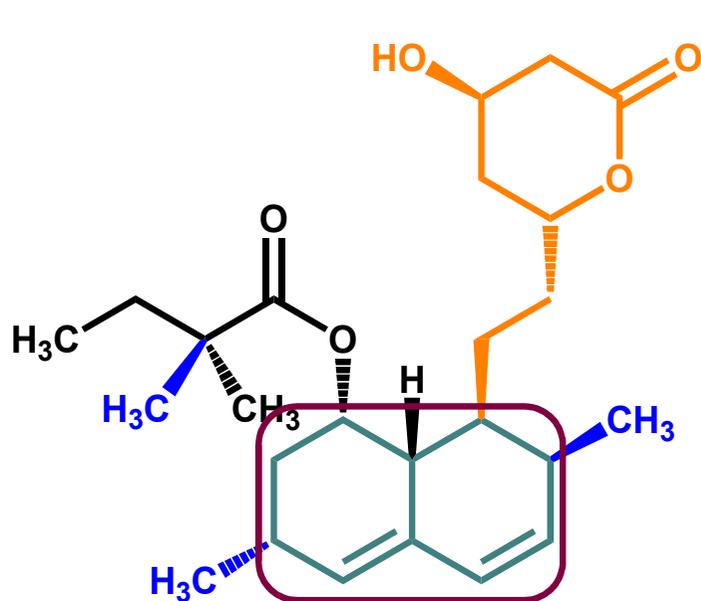
Imatinibe world sales in 2009: US\$ 4,0 bi*

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* S Aggarwal, Nature Rev Drug Discov 2010, 9, 427

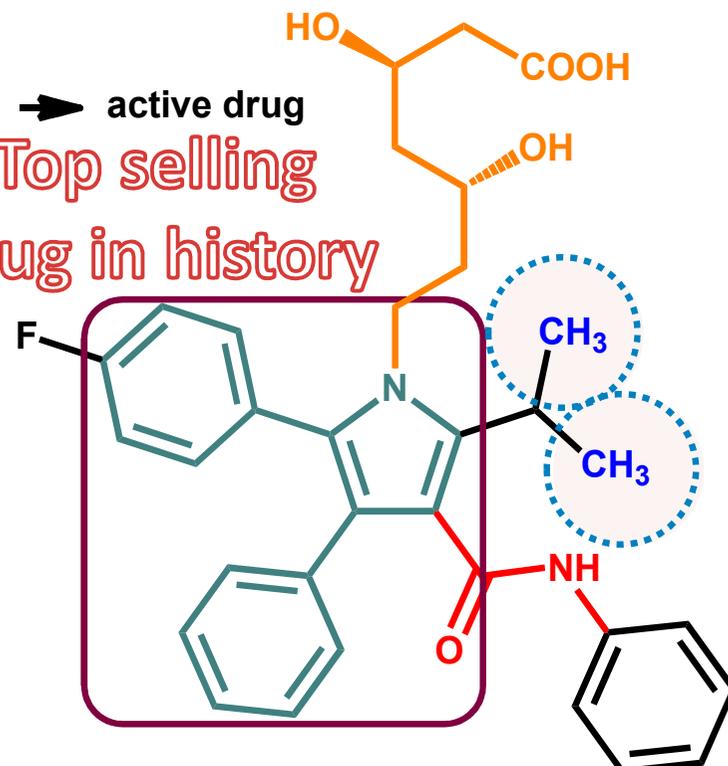


As metilinhas bilionárias...



simvastatin

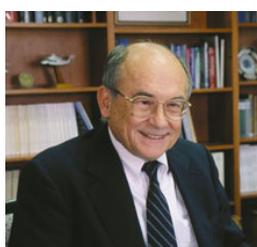
prodrug-like → active drug
Top selling drug in history



atorvastatin



A Endo



A A Patchett



B D Roth

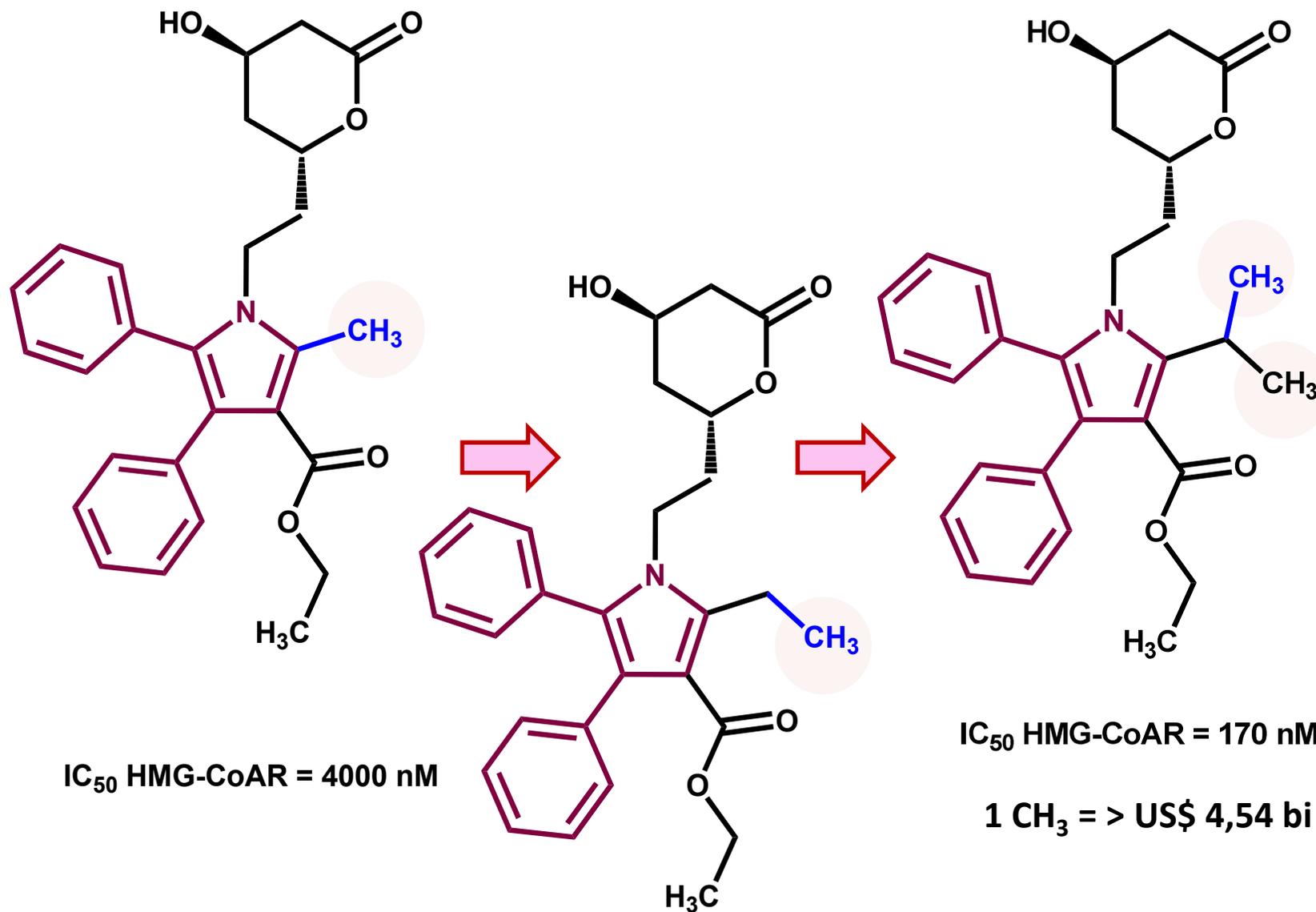
BD Roth, *Prog Med Chem* 2002, 40, 1



Pfizer *blockbuster* => **US\$ 150 bi**
 1991-2011



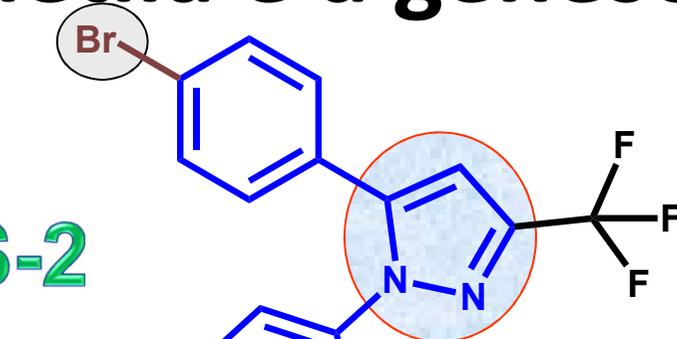
IC₅₀ HMG-CoAR = 8 nM





A metila e a gênese do celecoxibe

PGHS-2



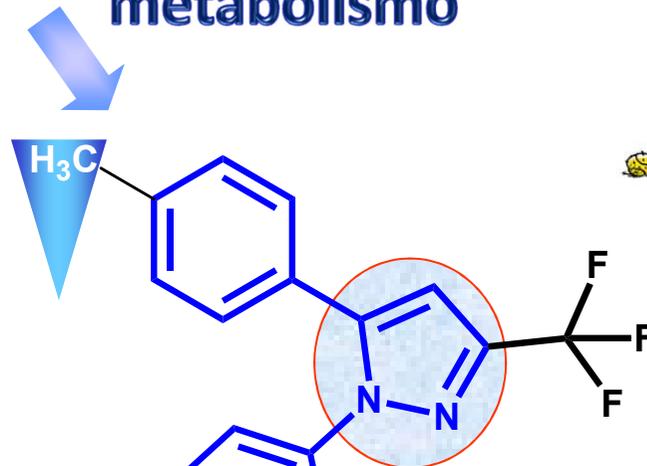
COX-2 seletivo
Searle



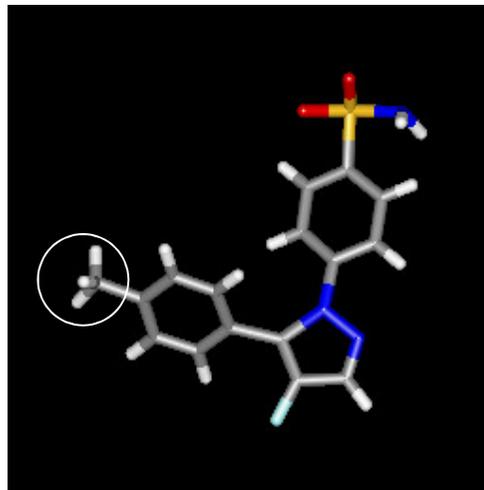
Vida-média = **12 dias!**
(ADME)



metabolismo



nova possível indicação:
câncer coloretal



Celecoxibe (SC-58634)

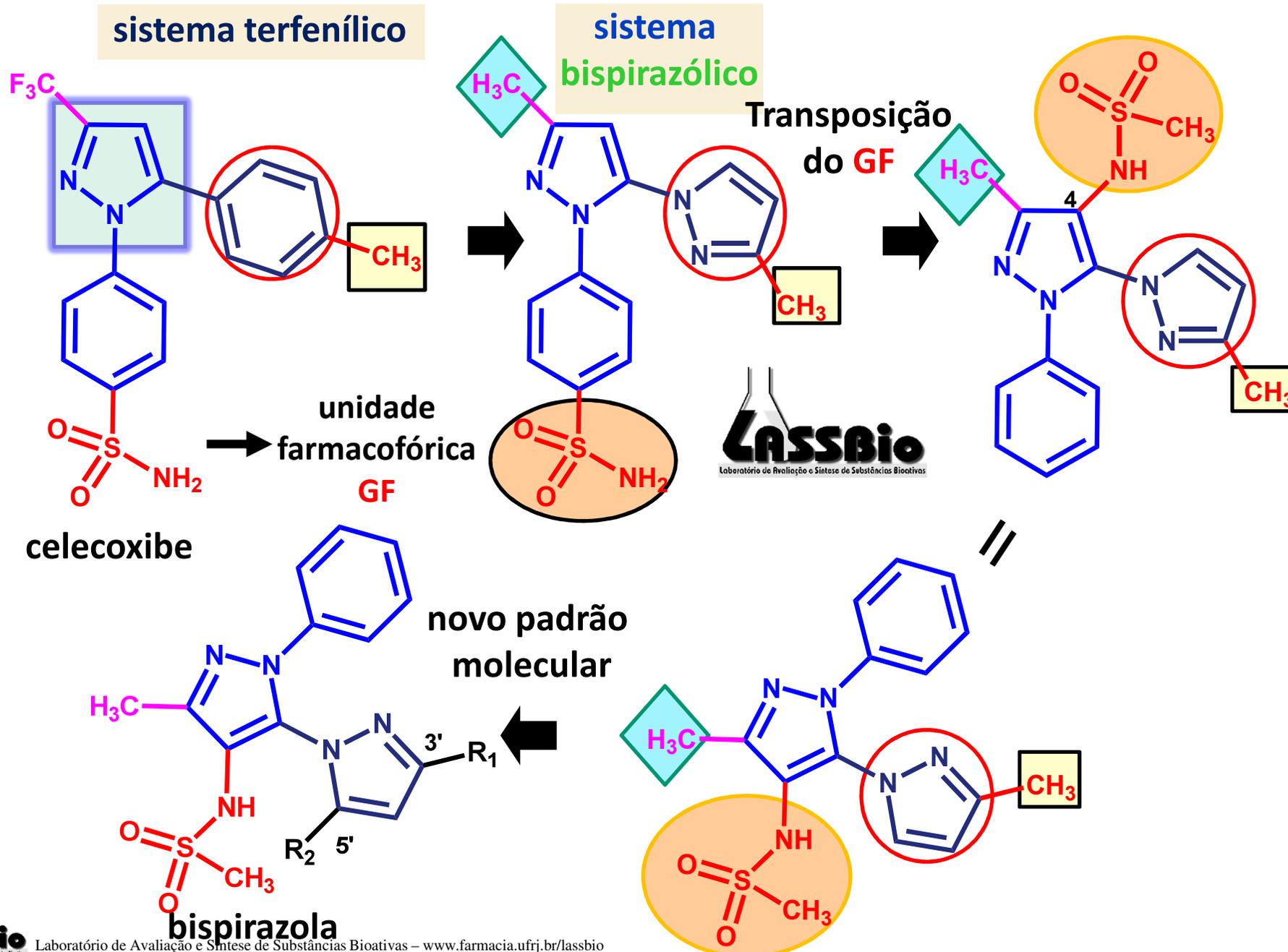
1999

TD Penning *et al.*, *J. Med. Chem.* **1997**, *40*,1347

O mercado mundial de fármacos antiinflamatórios (2008) = ca. US\$ 32 bi

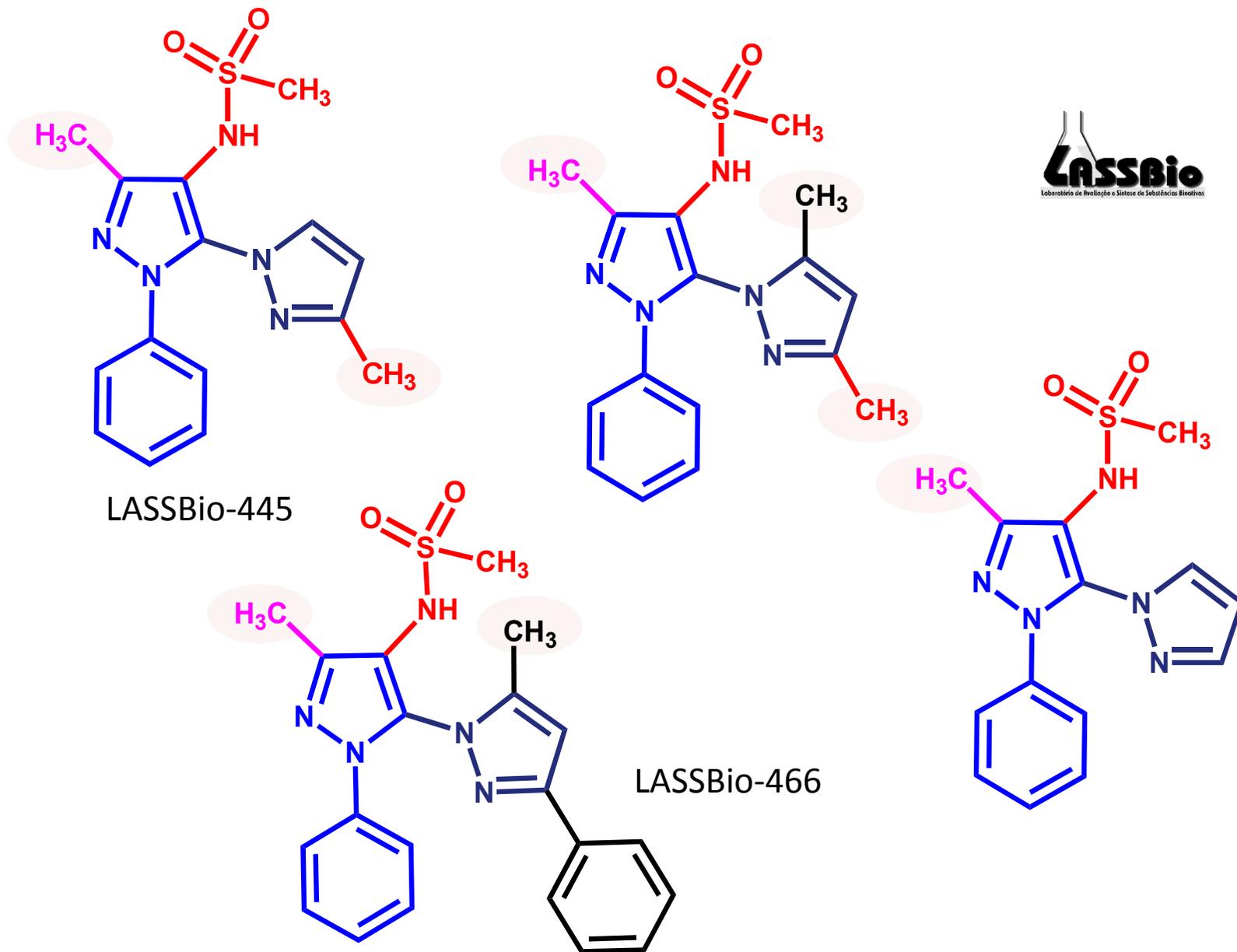


Desenho estrutural de novos COX-2i bispirazólicos



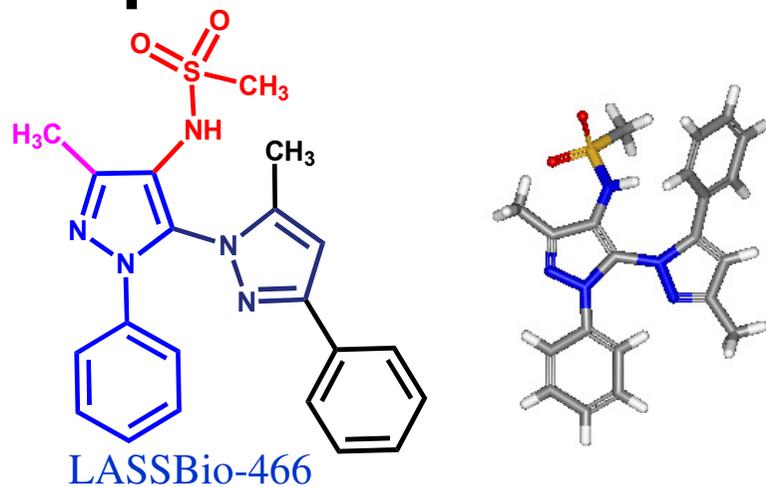


Série congênere dos novos COX-2i bispirazólicos

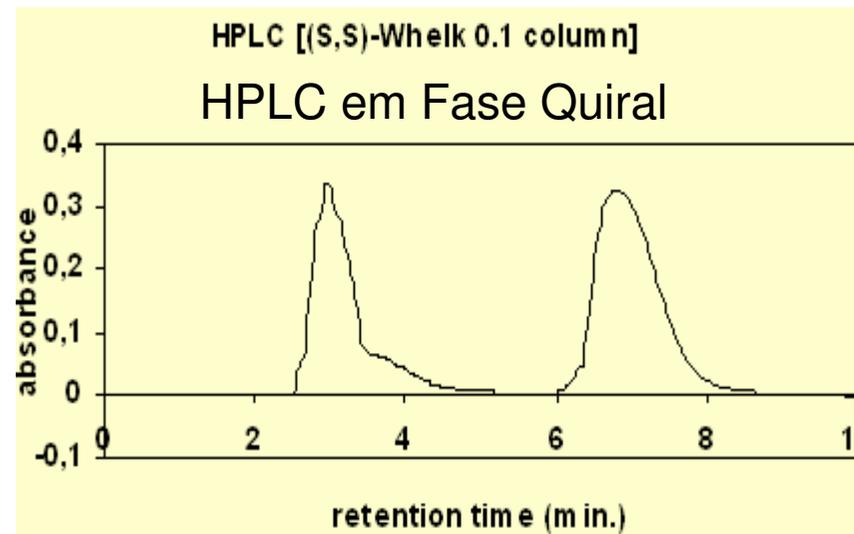
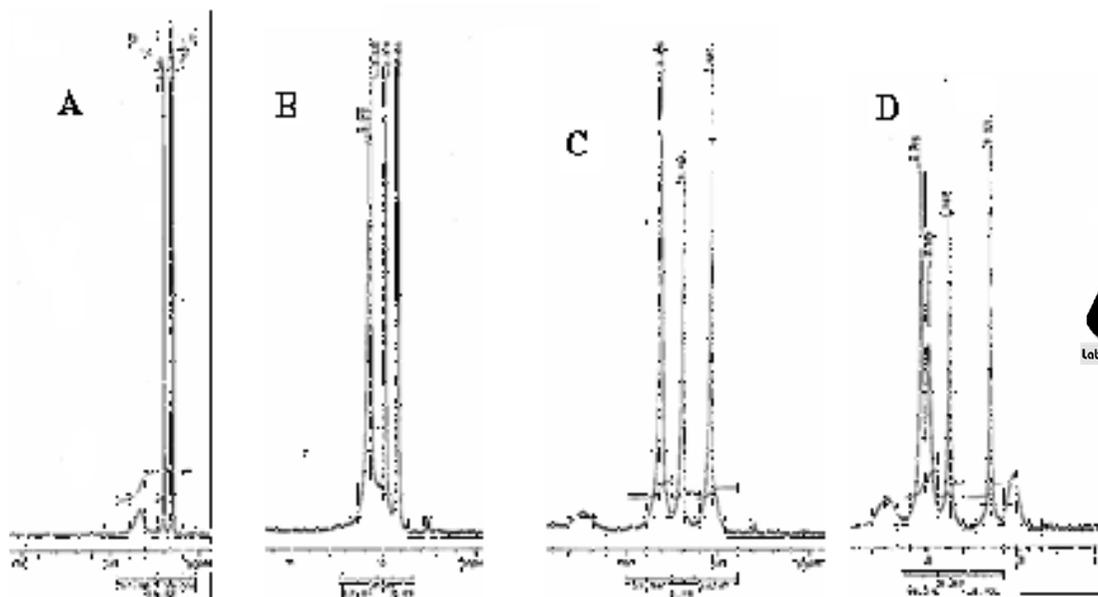




Atropoisomerismo de Derivados Bispirazólicos



Espectro da $^1\text{H-NMR}$ na presença de concentração variável de $\text{Yb}(\text{thc})_3$: A = sem adição; B = adição 1; C = adição 2; D = adição 3 (concentrações crescentes)



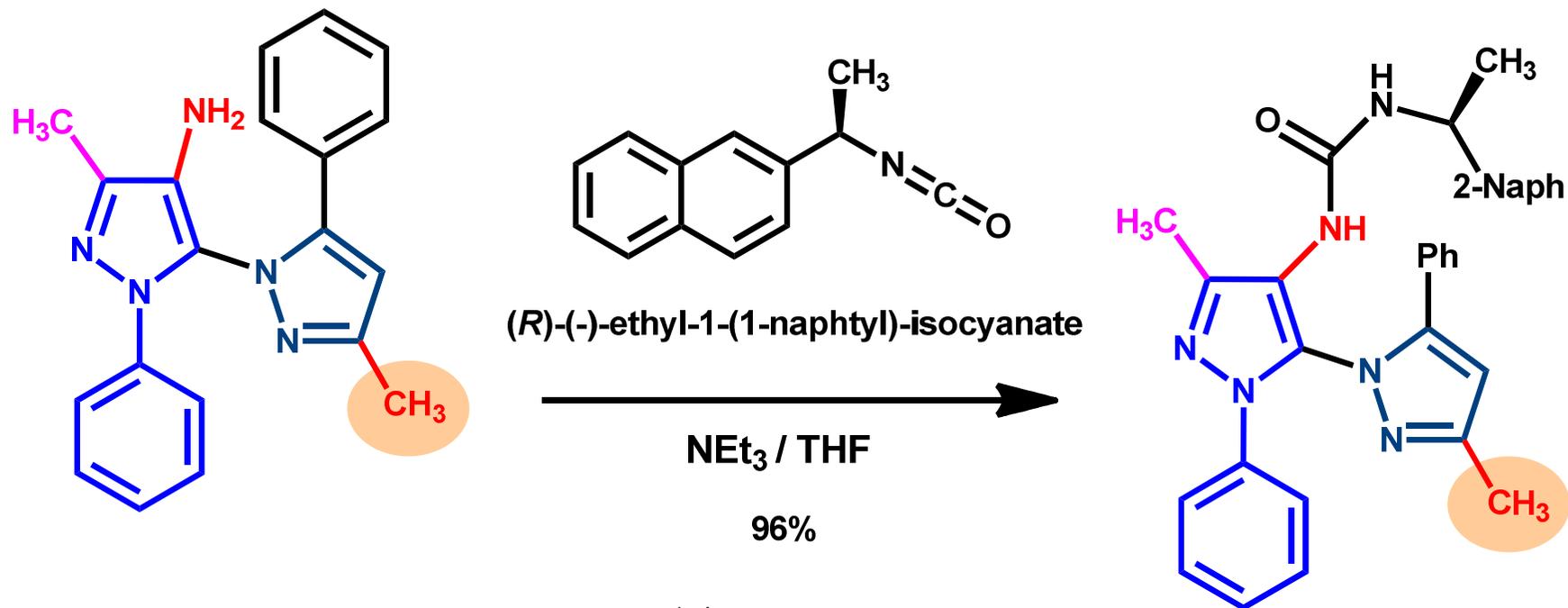
Prof. Helio M. Alves (FF-UFRJ)



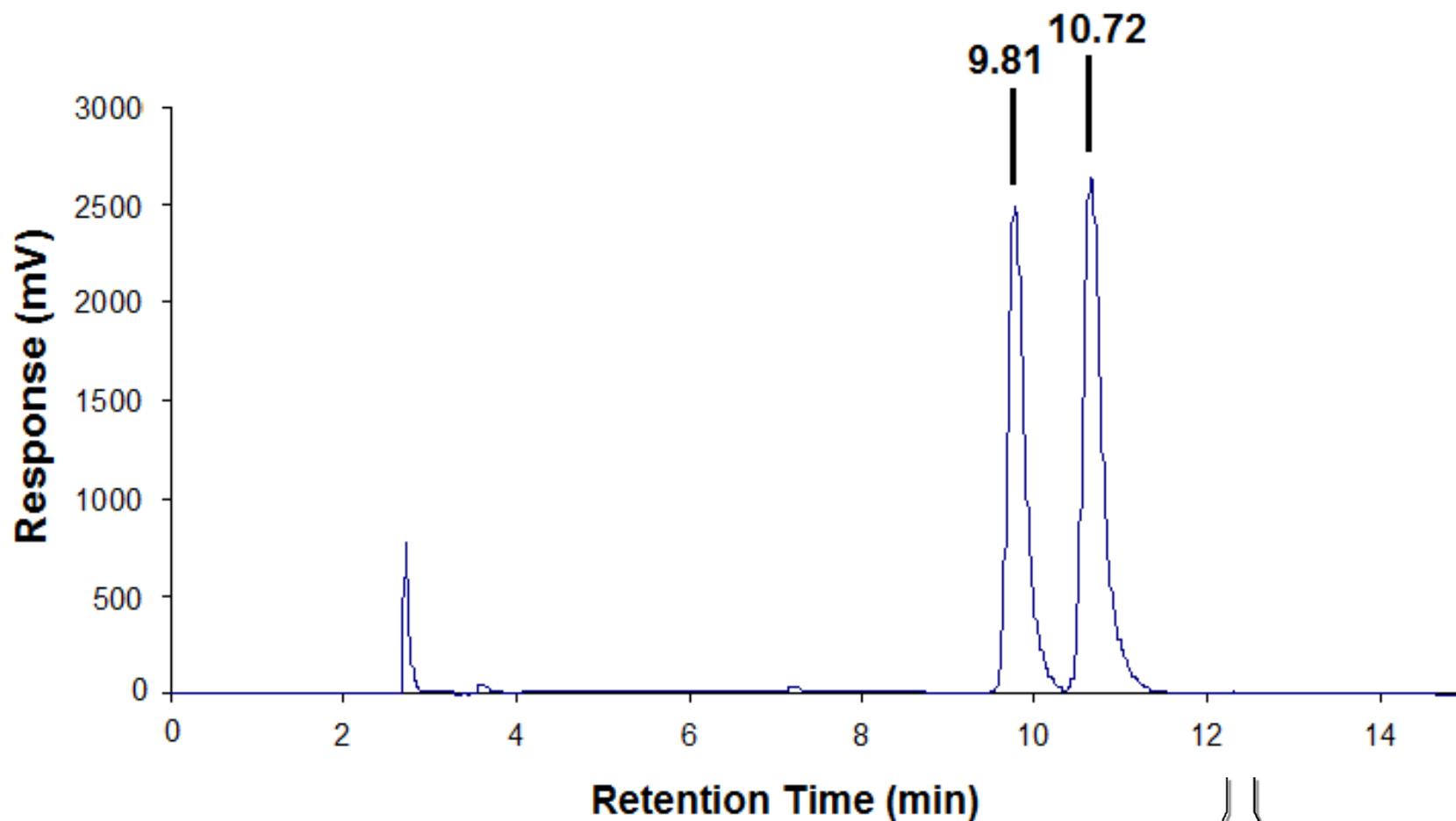
Veloso, M. P. (2000) Tese de Doutorado, IQ-UFRJ

E. J. Barreiro *et al.* (1999) INPI PI9902960-0

Derivatização a partir da amina



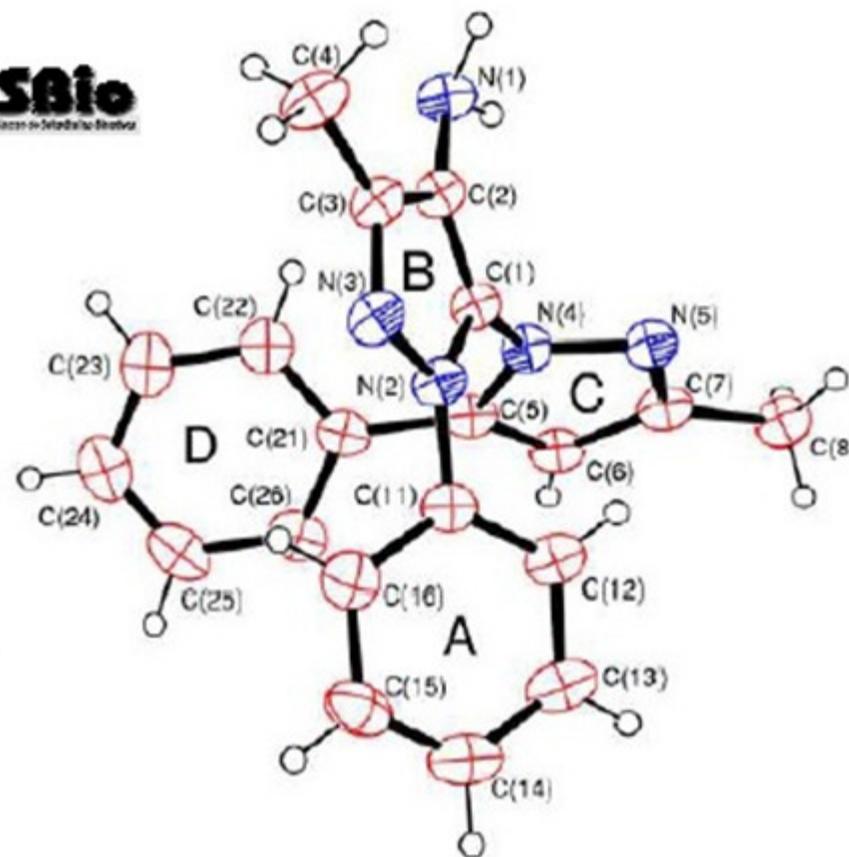
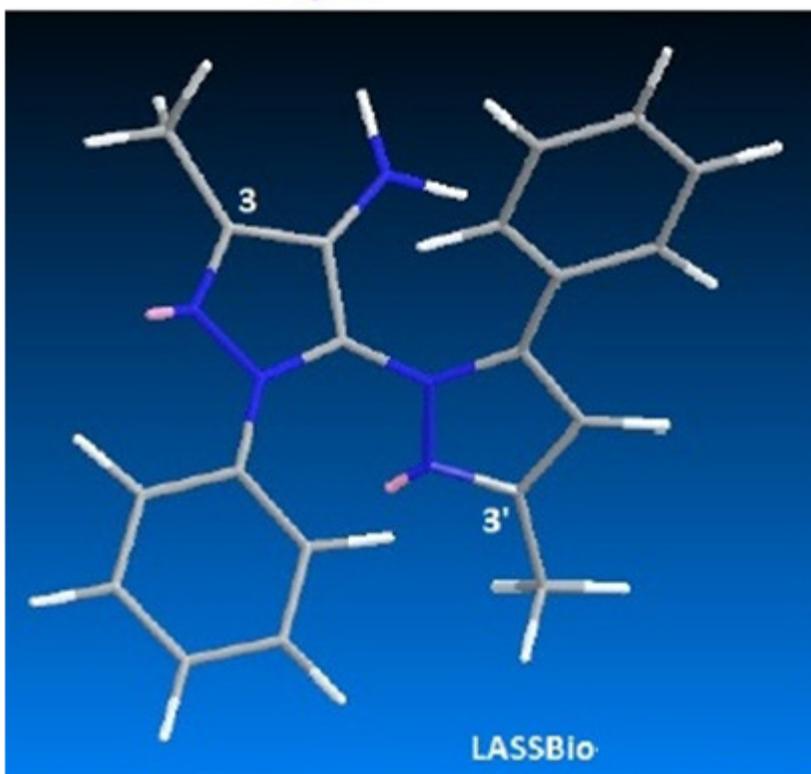
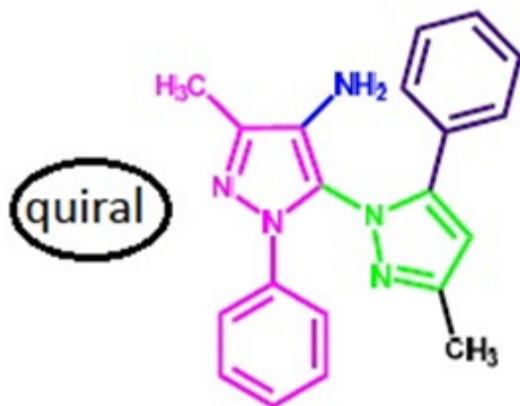
LASSBio-775



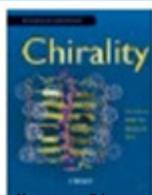
Lichosorb (N. 738342) RP-18 column (250 mm x 4 mm x 5 μ m)
L-7450A diode array detector (DAD)
acetonitrile and water (adjusted to pH 3 with TFA 0.1%) gradients



Atropoisomerismo em bispirazóis antiinflamatórios



ORTEP view of *P*-atropisomer



MP Veloso *et al.*, Synthesis and characterization of the atropisomeric relationships of a substituted *N*-phenyl-bipyrazole derivative with Antiinflammatory properties, *Chirality* **2012**, 00, 000.

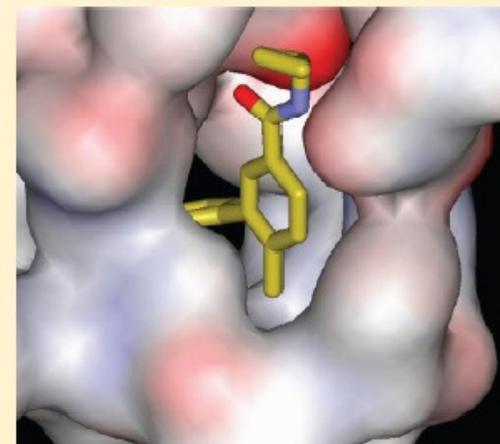


Methyl Effects on Protein–Ligand Binding

Cheryl S. Leung, Siegfried S. F. Leung, Julian Tirado-Rives, and William L. Jorgensen*
Department of Chemistry, Yale University, New Haven, Connecticut 06520, United States

J. Med. Chem. **2012**, *55*, 4489

ABSTRACT: The effects of addition of a methyl group to a lead compound on biological activity are examined. A literature analysis of >2000 cases reveals that an activity boost of a factor of 10 or more is found with an 8% frequency, and a 100-fold boost is a 1 in 200 event. Four cases in the latter category are analyzed in depth to elucidate any unusual aspects of the protein–ligand binding, distribution of water molecules, and changes in conformational energetics. The analyses include Monte Carlo/free-energy perturbation (MC/FEP) calculations for methyl replacements in inhibitor series for p38 α MAP kinase, ACK1, PTP1B, and thrombin. Methyl substitutions *ortho* to an aryl ring can be particularly effective at improving activity by inducing a propitious conformational change. The greatest improvements in activity arise from coupling the conformational gain with the burial of the methyl group in a hydrophobic region of the protein.



The importance of methyl groups in modulating biological activity for small molecules is well documented.¹ Consistent with this, the most fundamental change in structure–activity studies is replacement of a hydrogen atom by a methyl group.

(1) Barreiro, E. J.; Kummerle, A. E.; Fraga, C. A. M. The Methylation Effect in Medicinal Chemistry. *Chem. Rev.* **2011**, *111*, 5215–5246.



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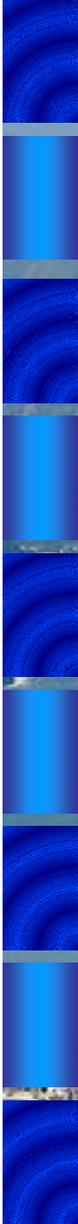
28 de janeiro a 01 de fevereiro

INSCRIÇÕES:
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